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Deep denitrification: stream and groundwater biogeochemistry

reveal contrasted but connected worlds above and below

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- 25 Key words: Nitrogen, Sulfur, Silica, Carbon, Groundwater, Surface Water, Denitrification, Environmental
- 26 Tracers, Residence Time, Nitrate Removal, Wells

Key points 27

- 1. Disparate processes create opposite chemical patterns in surface and subsurface 28
- 29 2. C, N, and S patterns reflect both mixing and separation at watershed scales
- 30 3. Groundwater is as or more dynamic than surface water for many parameters
- 4. δ^{34} S, DSi and CFCs provide robust nutrient retention and residence time proxies 31

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33 Abstract

34 Excess nutrients from agricultural and urban development have created a cascade of ecological crises 35 around the globe. Nutrient pollution has triggered eutrophication in most freshwater and coastal ecosystems, contributing to a loss in biodiversity, harm to human health, and trillions in economic 36 damage every year. Much of the research conducted on nutrient transport and retention has focused on 37 surface environments, which are both easy to access and biologically active. However, surface 38 39 characteristics of watersheds, such as land use and network configuration, often do not explain the 40 variation in nutrient retention observed in rivers, lakes, and estuaries. Recent research suggests subsurface processes and characteristics may be more important than previously thought in determining watershed-41 42 level nutrient fluxes and removal. In a small watershed in western France, we used a multi-tracer 43 approach to compare surface and subsurface nitrate dynamics at commensurate spatiotemporal scales. We 44 combined 3-D hydrological modeling with a rich biogeochemical dataset from 20 wells and 15 stream 45 locations. Water chemistry in the surface and subsurface showed high temporal variability, but 46 groundwater was substantially more spatially variable, attributable to long transport times (10–60 years) 47 and patchy distribution of the iron and sulfur electron donors fueling autotrophic denitrification. Isotopes 48 of nitrate and sulfate revealed fundamentally different processes dominating the surface (heterotrophic 49 denitrification and sulfate reduction) and subsurface (autotrophic denitrification and sulfate production). Agricultural land use was associated with elevated nitrate in surface water, but subsurface nitrate 50 51 concentration was decoupled from land use. Dissolved silica and sulfate are affordable tracers of 52 residence time and nitrogen removal that are relatively stable in surface and subsurface environments. Together, these findings reveal distinct but adjacent and connected biogeochemical worlds in the surface 53 54 and subsurface. Characterizing how these worlds are linked and decoupled is critical to meeting water 55 quality targets and addressing water issues in the Anthropocene.

57 **1 Introduction**

Excess nutrients from human activities have created a cascade of ecological crises that 58 59 threaten humankind and ecosystems around the globe. The nutrients that have constrained primary productivity and heterotrophic growth for most of Earth's history-nitrogen (N) and 60 phosphorus (P)—now saturate many environments along the terrestrial-aquatic-marine gradient, 61 62 including soils, rivers, lakes, groundwater, and oceans (Pinay et al 2018, Van Meter et al 2018, Kolbe et al 2019, Jane et al 2021, Dai et al 2023, Davidson et al 2023). This widespread nutrient 63 64 overload (eutrophication) causes immense damage to ecological food webs and human society, making it one of the most pressing and costly environmental issues of our day (Diaz and 65 Rosenberg 2008, Steffen et al 2015b, Le Moal et al 2019, Ward et al 2018). Globally, 66 approximately 80% of freshwater and coastal ecosystems are experiencing some level of 67 anthropogenic eutrophication (Galloway et al 2003, Frei et al 2020, Stoddard et al 2016, 68 69 Brahney et al 2015, Dai et al 2023).

In response to this nutrient crisis, progress has been made in some regions to stem the 70 flood of anthropogenic nutrients, largely through the control of nutrient point sources such as 71 72 wastewater discharge from humans and livestock (Abbott et al 2018b, Ehrhardt et al 2019, Yang et al 2019). Additionally, many countries have implemented practical and regulatory measures to 73 reduce nutrient pollution, though the results remain mixed (Poisvert et al 2017, Dupas et al 2018, 74 75 Van Meter et al 2018, Sabo et al 2021b). However, as human activities now directly affect 77% of the ice-free land surface (Watson et al 2018, Abbott et al 2019a), nutrient concentrations and 76 fluxes continue to rise in much of the world (Seitzinger et al 2010, Bouwman et al 2013, 77 78 Stoddard et al 2016, Le Moal et al 2019, Frei et al 2021).

To identify effective solutions to the eutrophication crisis, we need to understand the 79 sources and fates of nutrients added by humans to the Earth system. For example, agricultural 80 activities account for approximately 85% of the human nutrient load globally (Smil 1999, 81 Galloway et al 2004, Bouwman et al 2009, Campbell et al 2017). However, only half of the 82 nutrients applied in fertilizers is taken up by crops (Smil 1999, Liu et al 2010, Foley et al 2011), 83 84 with most of the nutrient waste occurring in areas of intensive agriculture in Asia, Europe, and North America (Foley et al 2011, Sabo et al 2021b). In agricultural areas where nutrient inputs 85 exceed nutrient uptake of harvested crops, there are three general fates for excess nutrients: 86 accumulation in the soil, leaching into groundwater or surface water, or conversion to gaseous 87 forms via denitrification for N (Aquilina et al 2012a, Sebilo et al 2013, Van Meter et al 2016, 88 Poisvert et al 2017, Minaudo et al 2019, Peterjohn and Correll 1984). While understanding the 89 relative importance of these fates in different socio-ecological contexts is crucial to solving 90 eutrophication, it is also exceedingly difficult at medium to large scales. On one end of this 91 92 scaling problem, plot-scale experiments in a single component of the watershed (e.g., the riparian zone or soil layer) are often unrelated to watershed-level fluxes (Pinay et al 2015, 2018, Heiner 93 et al 2022, Ebeling et al 2021, Basu et al 2022). On the other end, watershed-scale observations 94 95 often do not have the spatial resolution to identify what characteristics or practices are effective at reducing nutrient waste (Abbott et al 2018b, Thomas and Abbott 2018, Frei et al 2020, Van 96 97 Meter *et al* 2021).

While most interventions to improve water quality are associated with surface or nearsurface environments (e.g. riparian zones, surface waters, and soils), there is growing evidence
that catchment-level nutrient retention is strongly influenced by subsurface characteristics
(Aquilina *et al* 2018, Kolbe *et al* 2019, Frei *et al* 2020, Pauwels *et al* 2000, Böhlke 2002, Duncan

102	et al 2015, Fan et al 2020, Nguyen et al 2022a, Lupon et al 2023). Because hydrological
103	residence time in the subsurface far exceeds the surface (Gleeson et al 2016, Cook and Herczeg
104	2012, Green et al 2011), time lags or "legacies" are created between the input and output of
105	excess nutrients (Van Meter and Basu 2017, Ehrhardt et al 2019, Guillaumot et al 2021, Basu et
106	al 2022). This provides extended opportunities for physical and biological processes to take up
107	or transform nutrients and other solutes traveling through aquifers (Oldham et al 2013a, Abbott
108	et al 2016, Dehaspe et al 2021). However, subsurface environments have long been considered
109	as less dynamic than surface environments (Li et al 2017), where rapid biogeochemical
110	processes are easily observable (McClain et al 2003, Bernhardt et al 2017). This paradigm of a
111	"quiet world beneath" has recently been challenged on multiple fronts.
112	Advances in hydrological modeling, inferential methods, and direct observations now
113	show that the subsurface is biogeochemically active and temporally dynamic (Ben Maamar et al
114	2015, Long et al 2016, Marçais et al 2018, Kolbe et al 2019, Bochet et al 2020). However, the
115	relative influence of subsurface and surface activity on watershed-scale nutrient dynamics
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	remains uncertain. Because of logistical challenges and disciplinary divides (Li et al 2017, Kolbe
117	remains uncertain. Because of logistical challenges and disciplinary divides (Li <i>et al</i> 2017, Kolbe <i>et al</i> 2019, Krause <i>et al</i> 2022), relatively few studies characterize biogeochemistry in both the
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118 119 120	<i>et al</i> 2019, Krause <i>et al</i> 2022), relatively few studies characterize biogeochemistry in both the surface and deep subsurface (e.g., >10 m) at watershed scales, complicating direct comparison of the spatiotemporal dynamics and dominant processes in these environments. Yet, there is evidence from the shallow subsurface that these interactions are important across scales (Krause

In this context, we collected water samples over several years from streams and aquifers
in small watersheds (<50 km²) in a region of intensive agriculture in western France. Using a

multi-proxy framework combined with previous hydrological research from this area (Abbott et 125 al 2016, Kolbe et al 2016, Marcais et al 2018, Thomas et al 2019), we investigated the 126 interaction of hydrological residence time, land-use, and biogeochemical processing in surface 127 and subsurface environments. We were motivated by three general questions. First, how do 128 spatial and temporal variability regimes differ between the surface and subsurface environments 129 130 in this region? Second, what processes regulate nutrient dynamics as water travels through various watershed components (i.e., streams, vadose zones, and aquifers)? Third, what are the 131 management implications of these observations of watershed-scale ecohydrology in the surface 132 and subsurface? To address these questions, we focused on the transport and transformation of 133 nitrate (NO_3^{-}), one of the most common nutrients associated with eutrophication and one of the 134 costliest regarding environmental remediation, particularly in this region (Thomas et al 2019, 135 Abbott et al 2018b, Heiner et al 2022, Dupas et al 2018). We analyzed a broad range of elements 136 (e.g., carbon, sulfur, and silicon), proxies (e.g., isotopes, dissolved gases, and rare earth 137 138 elements), and other data (e.g., historical land use, 3-D hydrological modeling, and reconstructed nutrient inputs) to characterize the surface and subsurface at the watershed scale. 139

140 <u>2 Methods</u>

141 2.1 Site description and experimental design

142 This study was conducted within the Zone Atelier Armorique research area, which is part

143 of the Long-term socio-ecological research (LTSER) network (Thomas et al 2019). The history,

- socioecological characteristics, and available data from this LTSER are described in detail by
- 145 Thomas *et al* (2019). The research area is in northeast Brittany in northwestern France. Brittany
- has a temperate oceanic climate (category "Cfb": temperate, no dry season, warm summer,
- 147 following Kottek *et al* 2006) and extensive livestock and row-crop agriculture (Thomas *et al*

2019, Frei et al 2020, Poisvert et al 2017). The study catchment occurs on the Armorican massif, 148 which is composed of metamorphic and igneous rock, primarily granite, schist, and micaschist 149 (Aquilina et al 2012a, Goderniaux et al 2013b, Kolbe et al 2016). The mean annual temperature 150 is 11.2°C, and the mean annual rainfall is 910 mm, which is relatively well distributed 151 throughout the year, and mean potential evapotranspiration is 690 mm (Thomas et al 2019). 152 153 Because of the high infiltration capacity and relatively gentle precipitation, there is little artificial drainage (e.g., tile drains), but there are ditches and hedgerows around many fields, some of 154 which have been present since Medieval times (Thomas and Abbott 2018, Baudry et al 2000, 155 156 Forman and Baudry 1984). The main land use is agriculture, with around 80% of the region covered by row crops, pastureland, and indoor animal husbandry (pigs, poultry, and cows). This 157 gives Brittany one of the highest densities of livestock in France and Europe (Gascuel-Odoux et 158 al 2010, Poisvert et al 2017, Kim et al 2019). The intensive agriculture in this region has created 159 widespread and persistent eutrophication of streams, estuaries, and groundwater (Dupas et al 160 2018, Moatar et al 2017, Minaudo et al 2019). However, improved nutrient management, 161 particularly the reduction of point sources (e.g., wastewater and feedlot effluent) and decreases in 162 fertilizer application have resulted in declining N and P concentrations in many Brittany 163 164 watersheds (Dupas et al 2018, Poisvert et al 2017, Aquilina et al 2012b, Gu et al 2021, Abbott et al 2018b). 165

To characterize the dominant hydrological and biogeochemical processes in the surface and subsurface, we collected samples for chemical analysis from 16 agricultural wells (28-94 m deep; hereafter *deep wells*), 4 research piezometers (<10 m deep; hereafter *shallow wells*), and 15 surface water locations within Le Guyoult and the Couesnon watersheds (Fig. 1). From December 2014 to April 2016, we performed seven sampling campaigns to capture temporal

variability in the surface and subsurface. To maximize the spatial extent of surface and
groundwater sampling, not all the sites were visited during every campaign, and the mean
number of return visits per site was four.

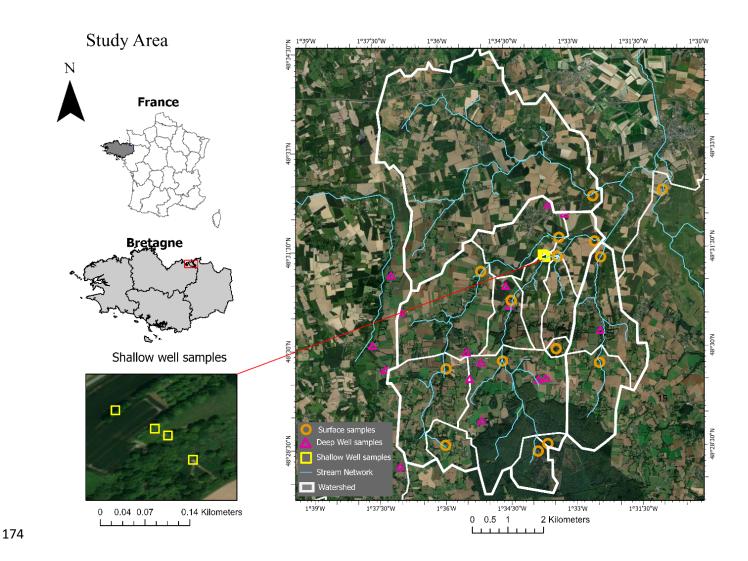


Figure 1. Map of the study area near the Zone Atelier Armorique Long-term socioecological research (LTSER) site. Le Guyoult watershed to the west is a tributary of the
larger Couesnon watershed to the east. The white lines indicate the subcatchment
boundaries based on topographic delineation.

180 **2.2 Estimating nitrate removal**

Nitrate removal was estimated based on the concentration of excess N2 dissolved in the 181 182 groundwater. The detailed procedure for calculating nitrate removal is described fully in Kolbe et 183 al 2019 and Vautier et al 2021. Briefly, in a groundwater environment, the concentration of excess N₂-i.e., above the expected concentration—can indicate the amount of NO₃⁻ removed by 184 185 denitrification because equilibration with the atmosphere is not possible (Aeschbach-Hertig et al 1999, Ayraud et al 2006, Böhlke 2002). After correcting for atmospheric partial pressure and 186 temperature with the observed Ar concentration, the sum of the excess N2 and remaining NO3-187 allow calculation of the concentration of NO₃⁻ that reached the water table (Kolbe et al 2019, 188 Böhlke 2002). Combined with estimates of groundwater residence time, this can allow the 189 reconstruction historical NO₃⁻ inputs (Kolbe *et al* 2019, Aquilina *et al* 2012b). 190

Though land use in specific parcels of the catchment has varied substantially in the past 191 (Barbe *et al* 2020), crop rotations have created a reasonably uniform N input time series at the 192 catchment scale (Kolbe et al 2019). Because this method only accounts for NO₃⁻ removed after 193 the water lost contact with the atmosphere (i.e., in the saturated zone), it accounts implicitly for 194 195 biogeochemical removal or retention in the unsaturated zone (Thomas and Abbott 2018, Basu et al 2022). For our purposes, this is an advantage over more common nutrient surplus estimates, 196 which account for overall disequilibrium in N and P but not vadose-zone removal (Van Meter et 197 198 al 2016, Sebilo et al 2013, Poisvert et al 2017).

199 **2.3 Proxies of hydrology and biogeochemistry**

To assess the degree of biogeochemical attenuation of nutrients and the relative
importance of nutrient loading versus nutrient removal, we measured the concentrations and

stable isotope ratios of NO₃⁻ (δ^{15} N and δ^{18} O) and SO₄²⁻ (δ^{34} S). Stable isotopes can indicate the 202 nutrient source and the degree of biogeochemical processing (Lehmann et al 2003, Mariotti et al 203 1981, Malone et al 2018). NO₃⁻ isotopes have been widely analyzed because NO₃⁻ is a dominant 204 form of nitrogen in nutrient saturated ecosystems (Aber et al 1998), organic and industrial 205 fertilizers have distinct initial δ^{15} N and δ^{18} O (Lohse *et al* 2013, Denk *et al* 2017, Bedard-Haughn 206 207 *et al* 2003), and denitrification (both heterotrophic and autotrophic) strongly fractionates NO_3^{-1} isotopes, increasing the residual δ^{15} N and δ^{18} O relative to the initial values prior to denitrification 208 (Hosono et al 2014, Malone et al 2018, Ayraud et al 2006, Pauwels et al 2010). Therefore, we 209 predicted that watersheds with isotopically enriched NO₃⁻ would have higher N attenuation 210 (Lehmann *et al* 2003) or alternatively that they would have received primarily organic fertilizer 211 (Bedard-Haughn et al 2003). 212

Sulfur isotopes have been used extensively in geochemical studies of processes ranging 213 from the evolution of the Earth's atmosphere to bacterial and archaeal dissimilatory SO₄²⁻ 214 reduction (Canfield 2001, Farguhar and Wing 2003). For the purposes of this study, we were 215 particularly interested in δ^{34} S, which can be increased by SO₄²⁻ reduction after depletion of NO₃⁻ 216 (e.g., in highly reduced wetland habitats) and decreased by SO_4^{2-} production in aquifers during 217 218 autotrophic denitrification (Hosono et al 2014, Abbott et al 2016). Consequently, we predicted that δ^{34} S of SO₄²⁻ would be higher in watersheds where surface processes dominate NO₃⁻ 219 removal, whereas δ^{34} S would be lower when groundwater processes were relatively more 220 important. 221

We used a combination of tracers to assess water residence time, including chlorofluorocarbons (CFCs), sulfur hexafluoride (SF6), and dissolved silica (DSi). CFCs and SF6 have been commonly used as tracers of water age due to their high detectability and low

reactivity in aqueous environments (Abbott et al 2016, Cook and Herczeg 2012, Busenberg and 225 Plummer 1992, Wilson and Mackay 1996). However, these anthropogenic gases cannot be used 226 to quantify the age of surface waters because they readily equilibrate with the atmosphere when 227 groundwater resurfaces. DSi has been shown to strongly correlate with subsurface residence time 228 (Marcais et al 2018, Becker 2013) and has been used to measure surface water ages in previous 229 230 studies in this area (Frei *et al* 2020). Though the concentration of DSi can be influenced by terrestrial and aquatic plant uptake in surface waters, we assumed conservative transport given 231 the short residence time and relatively high concentration of DSi in these headwater streams 232 (Marçais et al 2018, Delvaux et al 2013, Carey et al 2019, Abbott et al 2018b). 233

234 **2.4 Gas sampling and analysis**

CFC and SF6 measurements were taken at the Plateform Condate Eau from University of 235 Rennes 1. We collected water in steel ampoules (40 ml for CFC and 300 ml for SF6) by pumping 236 with a MP1 Grunfoss pump in the borehole. After stabilization, we measured the conductivity, 237 pH and dissolved O₂ using a custom sensor manifold preventing contact with the atmosphere 238 (Hach, Loveland CO, USA; model HQ440d multi; Fig. S1), and the ampoules were rinsed with 239 240 three times their volume of water and closed (Labasque et al 2014). CFC and SF6 concentrations were determined by a purge and trap extraction coupled to a GC/ECD (Busenberg and Plummer 241 1992, 2000, Labasque 2020). Uncertainties are around 3% for young groundwaters for CFC, 242 243 10% for SF6, and 20% for CFC on old (>50 years) groundwaters. The major dissolved gases (He, Ne, Ar, N₂, CO₂, CH₄) were sampled in 500 ml glass flasks and analyzed by headspace 244 extraction followed by µGC/TCD measurements, following (Sugisaki and Taki 1987). 245 246 Uncertainties are around 5% for Ne and He and 3% for other gases.

248 2.5 Solute and isotope collection and analysis

At each site, we collected a 5 L sample of water for immediate sensor readings and eight 249 250 smaller samples for laboratory analyses. For the laboratory analyses, we immediately filtered 251 subsamples using a 50 mL syringe and two 250 mL filter towers. We used a 0.2 µm cellulose acetate syringe filter to prepare samples for the analysis of cations, rare earth elements (REE), 252 253 and NO₃⁻ isotopes. Molybdate reactive phosphorus (MRP) concentration was determined colorimetrically via reaction with ammonium molybdate (Murphy and Riley, 1962), with a 254 precision of $\pm 4 \mu g l^{-1}$ (Gu *et al* 2018). Nitrate isotope samples were frozen immediately and 255 shipped to the UC Davis Stable Isotope Facility for analysis of δ^{15} N and δ^{18} O of NO₃⁻ by 256 bacterial denitrification assay (McIlvin and Casciotti 2011). Isotope ratios of δ^{15} N and δ^{18} O were 257 determined by measuring N and O using a ThermoFinnigan GasBench + PreCon trace gas 258 concentration system connected to a ThermoScientific Delta V Plus isotope-ratio mass 259 spectrometer (Bremen, Germany) with a precision of $\pm 0.4\%$ and 0.5% for δ^{15} N and δ^{18} O, 260 261 respectively. Cations were analyzed by inductively coupled plasma mass spectrometry (ICP-MS; Agilent 7700×, Santa Clara, USA, relative uncertainties \pm 5%). We quantified dissolved organic 262 carbon (DOC; Shimadzu TOC-5050A, Kyoto, Japan, precision \pm 5%), which is often considered 263 264 a master variable influencing physicochemical conditions in aqueous environments (Zarnetske et al 2018, Abbott et al 2016). Because DOC is the electron donor for both aerobic respiration and 265 heterotrophic denitrification (Hosono et al 2014, Sebilo et al 2019), high DOC can enhance NO₃⁻ 266 removal, and high NO₃⁻ can accelerate DOC oxidation (Abbott et al 2016, Kolbe et al 2019). 267 This two-way interaction contributes to the strong negative stoichiometric relationship between 268 NO₃⁻ and DOC observed in surface and groundwaters globally, including in this region (Taylor 269 and Townsend 2010, Frei et al 2020) 270

271	To determine SO ₄ ²⁻ isotopes, we filtered water samples on a Millipore TM system using
272	0.45 μ m cellulose acetate filters. Filtered solutions were then heated to 70°C and a 5% solution
273	of barium chlorate was added drop-to-drop to precipitate dissolved SO ₄ ²⁻ as barium sulfate
274	(BaSO ₄). Sulfur isotope composition of SO ₄ ²⁻ was measured using a VarioPYROcubeTM
275	elemental analyzer in combustion mode interfaced in continuous-flow mode with an $Isoprime^{TM}$
276	isotope ratio mass spectrometer at the Laboratoire de Géologie de Lyon (CNRS UMR 5276,
277	University Claude Bernard Lyon 1). BaSO ₄ from water samples was analyzed by weighing 3
278	aliquots of 250 μ g in tin foil capsules. Measurements were calibrated against the three BaSO ₄
279	international standards NBS127, IAEA-SO-5 and IAEA-SO-6. The standard deviation of $\delta^{34}S$
280	measurements was close to 0.3‰. Data are reported as δ^{34} S vs. V-CDT.

281 **2.6 Spatial analysis**

Using detailed land-use time series from the LTSER (Barbe et al 2020), we calculated the 282 dominant land use type for each parcel from 1993 to 2013 (except 1995; Fig. 1). We converted 283 all the land use shapefiles to raster files with a cell size of 15 m. We then used "majority" in cell 284 stat to calculate the value of the most common land use through time. For the cells that had no 285 286 historical data, we used the 2013 values (i.e., the most recent values). Similar land-use types were clustered into 4 groups: row crops, developed land, natural landscapes, and pastureland. 287 The row crops category included corn, wheat, and other ploughed crops. Developed land 288 289 included asphalt roads, unpaved roads, buildings, and railroad tracks. Natural landscapes consisted of forested and riparian areas. The last category pastureland consisted of grazing 290 pastures and abandoned lands. 291

We intersected the land use raster with contributing watershed area for stream locations and the contributing area of the deep wells, which had been calculated in previous work (Kolbe

et al 2016). We excluded the shallow wells from this analysis because all the sampling points
were contained in the same subcatchment and therefore could not be differentiated by land use
(see Thomas *et al* 2019 for a detailed analysis of the shallow wells).

297 2.7 Statistical analysis

We used Pearson and Spearman correlations to quantify relationships among and within 298 biogeochemical and landscape parameters (Malone et al 2018). We calculated both parametric 299 (r) and nonparametric (rho) correlation coefficients for different reasons. The Spearman 300 301 correlations (rho) are rank based, providing a robust metric of the overall association across sites. Conversely, the Pearson correlations (r) provide full weight to extreme values, which can be 302 303 influential in determining overall watershed behavior. For example, small portion of the watershed (i.e., a single or a few sites) can control source or sink dynamics for a given parameter 304 (Bernhardt et al 2017, McClain et al 2003, Abbott et al 2018a, Lee et al 2022). For both 305 Spearman and Pearson correlations, we used a decision criterion of $\alpha = 0.05$. To compare 306 variability in surface and subsurface environments, we calculated the mean and its standard error 307 of the spatial and temporal coefficients of variation (CV) for each parameter. This allowed us to 308 309 compare whether there was more variation for a given parameter in the surface or in the subsurface. We conducted all analysis and visualizations in the R statistical computing software 310 environment (R Core Team 2022). 311

312 <u>3 Results</u>

313 **3.1** Spatiotemporal dynamics in the surface and subsurface

The water chemistry of wells and streams indicated both contrasts and connectivity of the surface and subsurface water. The subsurface showed larger variations in most physicochemical

parameters compared to the surface sites, except for temperature, pH, and DOC, which are
strongly influenced by seasonal drivers at the surface (Fig. 2). For most parameters, including the
nutrients which are the focus of this study, the stream values fell between the deep and shallow
wells (Fig. 2). The N parameters grouped closer to deep wells, but the P, S, and Si parameters
grouped closer to shallow wells, suggesting a combination of mixing and biogeochemical
attenuation at watershed scales.

Separating the combined variability shown in Figure 2 into spatial and temporal CV 322 323 revealed several surprising patterns (Fig. 3). Throughout the watershed and through time, stream 324 chemistry showed relatively less variation, with no parameters exceeding a mean CV of 75% (Fig. 3). Conversely, the shallow wells and especially the deep wells showed greater spatial and 325 temporal variability, including several parameters with >100% variation (Fig. 3). We note that 326 CV is expressed as a percentage, representing proportional change rather than absolute change. 327 For example, stream environments showed more absolute range in DOC concentration (Fig. 2), 328 but because the mean DOC was much higher than in the subsurface sites, the DOC CV was 329 lower in streams (Fig. 3). 330

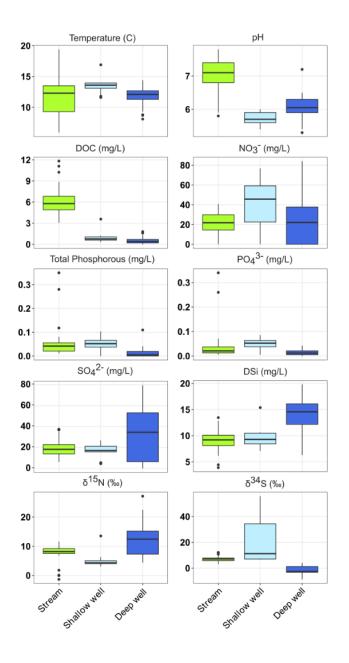


Figure 2. The distribution of several physicochemical parameters in the streams, shallow wells, and deep wells sampled in this study. The parameter and units are shown above each plot. The boxplots represent the median, interquartile range (IQR), points within 1.5-times the IQR, and outliers beyond 1.5-times the IQR. Because each site and sampling event are included as individual points, the ranges represent the combined spatial and temporal variability.

Spatial variation generally exceeded temporal variation across site types (i.e., most parameters plotted below the 1:1 line in Figure 3), though this was notably not the case for $NO_3^$ in deep wells and SO_4^{2-} and water age in all wells (see section 3.3 for detailed water age results). N and S isotopes tended to be substantially less variable than N and S concentrations, suggesting that hydrological dilution rather than changes in nutrient source or processing accounted for variability in concentrations.

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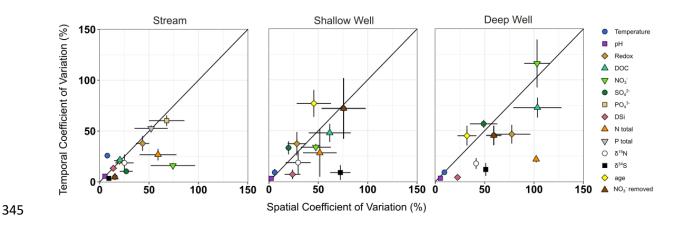
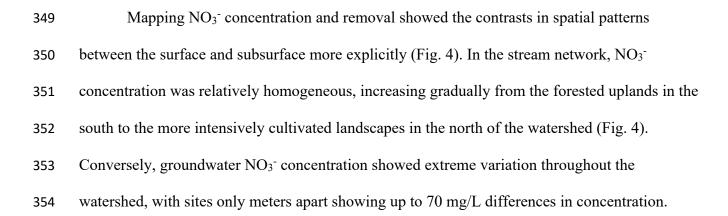


Figure 3. The spatial and temporal coefficients of variation for physicochemical parameters in streams, shallow wells, and deep wells (mean \pm SE).



Similarly, the percentage of NO_3^- removed was much more variable for groundwater, while the stream network showed a moderated pattern similar to the groundwater mean (Fig. 4).

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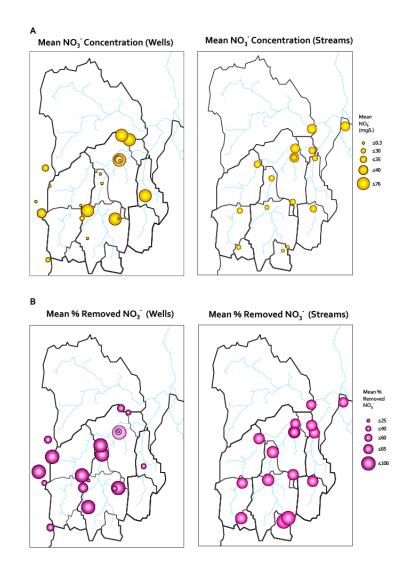
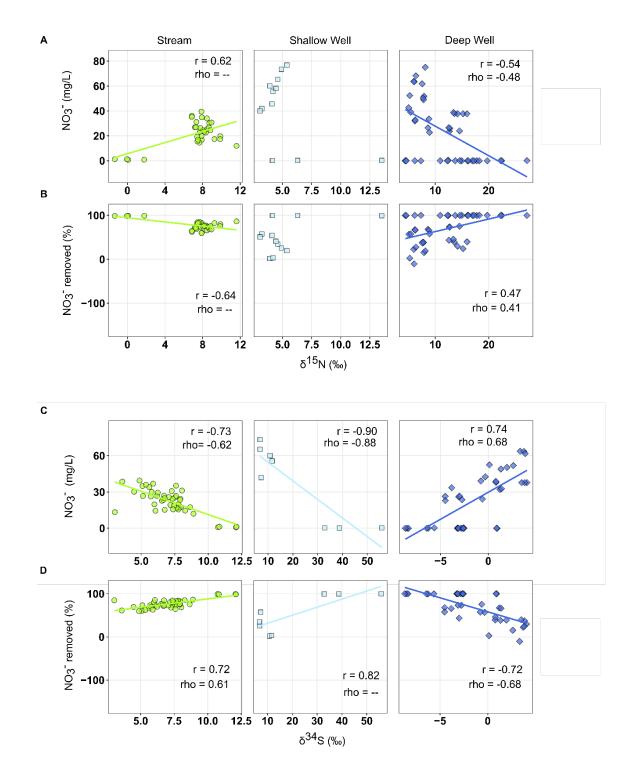


Figure 4. NO₃⁻ concentration (A) and percent removal (B) in subsurface (left) and
surface (right) environments. Average across time for each site, showing spatial variation
of chemical parameters. NO₃⁻ removal was calculated based on the difference between the
reconstructed NO₃⁻ input and the observed concentration (methods).

363 3.2 Isotopic and dissolved tracers

The isotopes and solutes provided clues about the causes of the surprising spatiotemporal 364 365 patterns. There were opposite relationships between isotope signatures and NO₃⁻ removal in the 366 streams and deep wells, with mixed patterns in the shallow wells (Fig. 5A & 5B). In the deep wells, NO₃⁻ removal was positively correlated with δ^{15} N, as expected due to fractionation during 367 368 denitrification. The opposite pattern existed in the surface water where $\delta^{15}N$ was negatively correlated with NO₃⁻ removal, though the correlation was only significant for the Pearson 369 coefficient, indicating that a few extreme values were driving the relationship (Fig. 5A & 5B). 370 Indeed, the low NO₃⁻ and low δ^{15} N stream sites occurred in the forested south of the watershed 371 (Fig. 4), likely indicating a non-fertilizer N source (e.g., atmospheric deposition or natural N 372 fixation). We note that the percentage N removal for these sites was likely overestimated because 373 our reconstructed N input assumed similar land use for each subcatchment, which is not the case 374 for forested sites that have never been under cultivation. 375 In general, δ^{34} S was a better proxy of NO₃⁻ concentration and removal, with δ^{15} N 376

showing fewer and weaker relationships (Fig. 5). δ^{34} S was positively correlated with NO₃⁻ removal in the surface but negatively correlated in the deep wells (Fig. 5C & 5D), in line with our hypotheses about S reduction in the surface and S release from autotrophic denitrification in the subsurface.



381

Figure 5. Relationships of δ^{15} N and δ^{34} S with NO₃⁻ concentration and removal. Linear fit lines are shown for convenience to indicate a significant relationship—either Pearson (r) or Spearman (rho)—at $\alpha = 0.05$.

Contrary to our predictions, DOC was not correlated with most NO₃⁻ and SO₄²⁻ parameters, and when correlations existed, they were typically weak (Fig. S2). Across site types, there was a significant negative correlation between δ^{34} S and SO₄²⁻ concentration (Fig. 6). Some of the highest δ^{34} S values we are aware of were observed in the shallow well sites that occurred in the riparian wetland (Fig. 1). Together these patterns suggest that different processes are controlling nutrient retention in the surface and subsurface, with heterotrophic denitrification and sulfur reduction dominating in the surface and autotrophic denitrification in the subsurface.

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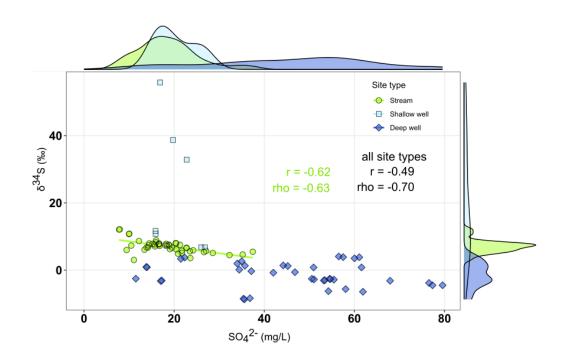


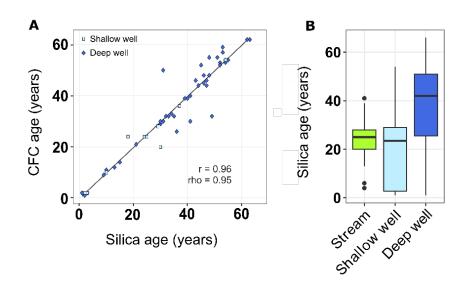
Figure 6. The relationship between δ^{34} S and SO₄²⁻ for streams, shallow wells, and deep wells. The two subsurface site types showed more variation in δ^{34} S, bracketing the stream water samples, which resembled a mix of the two contributing flow paths (i.e., deep groundwater and shallow soil/riparian water).

398

399 **3.3 Residence time and nutrient removal**

We combined gaseous and dissolved residence time tracers to allow estimation of mean water age in both the surface and subsurface. There was a high correlation (r = 0.96, rho = 0.95) between water residence times derived from DSi concentrations with residence times calculated using CFCs and SF6 (Fig. 7). This allowed us to use DSi to estimate mean residence time in surface water environments where gaseous tracers are not effective.

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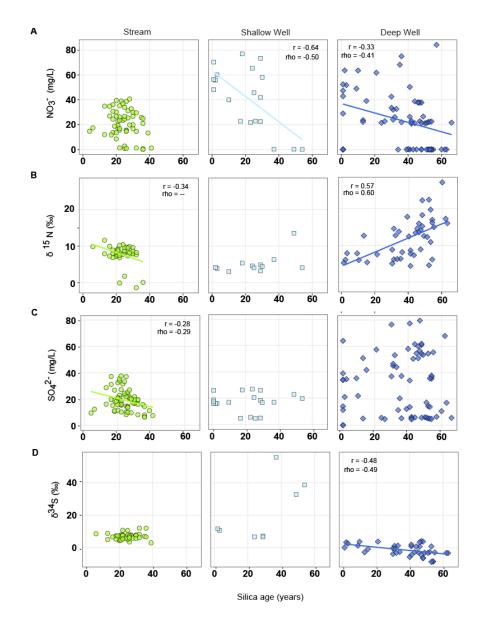


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Figure 7. Estimates of mean water age. Because the dissolved anthropogenic gas
 measurements are only effective in the subsurface, we calculated the relationship between
 silica-derived and CFC-derived water ages for shallow and deep groundwater
 environments only.

411

The deep wells showed the longest residence time, with a median age of 42 years and a mean age of 36 years. Shallow well water was the youngest, with a median age of 23.5 years and a mean age of 20.5 years (Fig. 7B). The stream water was slightly older than the shallow well
water with a median age of 25 years and a mean age of 24.4 years, indicating a mix of shallow
and deep groundwater sources. The streams were more homogeneous in age (interquartile range
of 20-28 years) than either the shallow (2-30 years) or deep wells (17-46 years; Fig. 7).

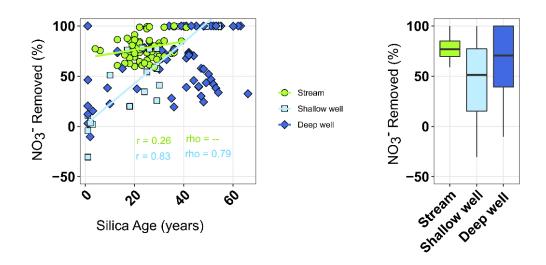


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Figure 8. Relationships between mean water age and SO₄²⁻ and NO₃⁻ isotopes and
 concentrations in stream, shallow groundwater, and deep groundwater environments.

422 NO_3 tended to be higher in younger water, i.e., there was a negative correlation between mean water age and NO₃⁻ concentration in shallow and deep wells (Fig. 8A). This implies that 423 either NO₃⁻ loading has increased through time or that older sites have less NO₃⁻ because of 424 cumulative removal. The latter is suggested by the positive correlation between water age and 425 δ^{15} N in deep wells, though the relationship is less clear in shallow wells and streams (Fig. 8B). 426 427 For streams and shallow wells, water age was positively correlated with NO₃⁻ removal (Fig. 9). We note that because the stream water comes from shallow and deep groundwater in this system, 428 the stream removal values are cumulative (i.e., most of the removal is happening in the deep and 429 430 shallow wells before flowing into the stream; Fig. 9).

431



432

433 Figure 9. NO_3^- removal versus mean water age for streams, shallow wells, and deep 434 wells. NO_3^- removal represents the cumulative removal along the flow path starting with 435 input of NO_3^- to the water table.

437 **3.4 Nutrient sources**

In stream environments, NO_3^- concentration was positively correlated with percentage of row crop agriculture in the subcatchment and negatively correlated with the amount of natural land cover types (Fig. 10). However, no such relationships were detected in deep wells, where NO_3^- concentration was not correlated with land use in the contributing area (Fig. 10). This lack of correlation could be due to a combination in uncertainty in the determination of contributing areas, temporal changes in land use over the ~40 years of transit time, and heterogeneous rates of removal of NO_3^- during transit through the aquifer.

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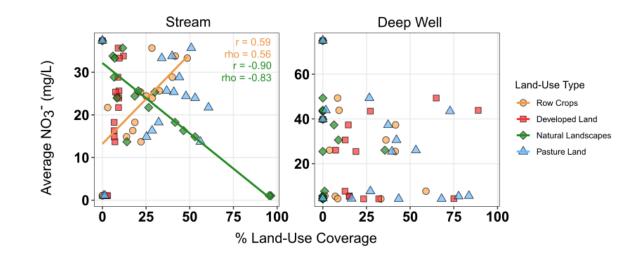


Figure 10. Correlation of NO_3^- and the percent of land-use cover for streams and deep wells. Shallow wells were omitted from this analysis as all shallow wells were contained within one subcatchment. Land-use coverage in the streams was calculated for the contributing subcatchment area. Land-use for deep wells was calculated from the well's contributing area based on prior hydrogeological modeling (Kolbe *et al* 2016).

452

453 <u>4 Discussion</u>

In this study we characterized biogeochemical and hydrological processes in the surface 454 455 and subsurface at the same spatiotemporal scales to explore watershed-level controls on nutrient 456 retention and removal. We used a multi-proxy approach to identify dominant NO₃⁻ removal processes and characterized spatiotemporal variation in streams and groundwater. We observed 457 458 highly distinct but linked biogeochemical "worlds" above and below the surface, which interacted to determine watershed-level nutrient dynamics. In the following paragraphs, we 459 460 compare these results with studies from this and other regions and explore potential management implications. 461

462 **4.1 Deepening our thinking to address water-energy-food crises**

Because of disciplinary differences in methods and processes, groundwater and surface 463 water environments are often considered individually, especially at watershed scales (Bochet et 464 al 2020, Abbott et al 2018a, Leibundgut et al 2009, Krause and Bronstert 2007, Duncan et al 465 2015, Knee et al 2018). Groundwater biogeochemistry and hydrology have long been areas of 466 focus in hydrogeology and critical zone science, but they have not received as much attention in 467 ecosystem ecology and nutrient management (Chorover et al 2007, Krause et al 2022, 468 469 Goderniaux et al 2013a, Li et al 2021, 2017, Brookfield et al 2021). Because attaining surface water quality goals depends on the overall watershed response (including the subsurface), this 470 disconnect needs to be addressed (Van Meter et al 2021, Vautier et al 2021, Heiner et al 2022, 471 472 Basu et al 2022, Lupon et al 2023). Additionally, global water security and aquatic habitat conservation increasingly depend on hydrochemical understanding of groundwater-surface water 473 interactions (Hartmann et al 2021, Jasechko and Perrone 2021, Wine 2022, Gleeson et al 2020). 474

Given how different the skillsets are for surface and subsurface measurements and 475 modeling, we believe the best way to deepen our watershed approaches is through 476 collaborations. We call on researchers and research-funding agencies to encourage such 477 collaboration. There are many watershed-focused projects from around the world that can be 478 used as models for next steps in integrating surface and groundwater hydrochemistry, ecology, 479 480 and sustainability in the Anthropocene (Oldham et al 2013b, Mu et al 2020, Krause et al 2022, Lee-Cullin et al 2018, Marcais et al 2022, Musacchio et al 2020). However, many watershed 481 482 observatories and interdisciplinary approaches are threatened or underfunded (Laudon et al 2017, Arènes et al 2018, Brooks et al 2015, Linton 2008, 2014, Thomas et al 2019). Increased 483 investment is needed not only for basic research but also (and perhaps primarily) for community 484 participation and education (Abbott et al 2018b, 2023, Ben-zvi-Assarf and Orion 2005, Basu et 485 al 2022). In the Anthropocene, the lack of understanding of watershed processes is a major 486 obstacle to achieving sustainable practices central to the water-energy-food nexus (Albrecht et al 487 2018, Zhang et al 2018, Biggs et al 2015, Abbott et al 2019b). 488

489 4.2 Integrated sampling reveals contrasted but connected worlds above and below

490 An important theme from collaborative work linking surface and subsurface 491 environments is that the world beneath is not stable or calm. Groundwater environments are commonly characterized as homogeneous and slow, lacking the diversity and energetic gradients 492 to support dynamic biogeochemistry (Bochet et al 2020, McClain et al 2003, Cardenas 2015). 493 494 Evidence from work around the world shows that the view of a quiet world beneath is not accurate (Marçais et al 2022, Ben Maamar et al 2015, Zhang et al 2015, Jasechko et al 2017, 495 Green et al 2011). Indeed, there is a growing consensus that catchment-level nutrient removal 496 may depend largely or primarily on subsurface characteristics that are often temporally dynamic 497

498 (Aquilina *et al* 2018, Kolbe *et al* 2019, Frei *et al* 2020, Bochet *et al* 2020, Long *et al* 2016,
499 Brookfield *et al* 2021, Hosono *et al* 2014).

For the watersheds in our study, the subsurface appears to be the epicenter of 500 denitrification rather than a biogeochemical afterthought, likely because of the combination of 501 longer residence times and relatively abundant electron donors (Oldham et al 2013b, Pu et al 502 2014). Groundwater environments are home to most of Earth's bacterial and archaeal biomass 503 504 (Bar-On et al 2018), and the combination of biological and hydrological factors can create 505 extreme spatial and temporal variation in groundwater reaction types and rates (Bochet *et al* 2020, Ben Maamar et al 2015). Because groundwater and surface-water environments are 506 507 constantly exchanging water, energy, and material, the failure to adequately characterize subsurface processes likely explains some of the difficulties in predicting nutrient fluxes and 508 recovery patterns at watershed scales (Kolbe et al 2019, Frei et al 2021, Van Meter et al 2018, 509 Brookfield et al 2021). 510

511 The relatively low variance we observed for many surface parameters demonstrates another interesting implication of surface-subsurface interactions. In Brittany, as in many 512 513 temperate regions, surface water is comprised of water mixing from the shallow and deep 514 groundwater (Marçais et al 2018, 2022, Thomas and Abbott 2018). While the processes and solute concentrations in the groundwater system are highly variable, they become mixed as they 515 enter the surface environment through riparian zones and diffuse groundwater flux to streams 516 517 (Krause et al 2022, Pinay and Haycock 2019, Le Moal et al 2019, Cardenas 2015). Given the short transit times in the surface, this results in a homogenized signature; as the surface water 518 519 travels farther and faster, it integrates a larger spatial scale and reflects a larger landscape patch (Abbott et al 2018a, Dupas et al 2019). While this reduces the ability of streams to mitigate high 520

521 nutrient loading (i.e., stream and riparian retention cannot offset watershed-wide nutrient

522 excess), it reinforces the role of streams as integrators of watershed processes, creating valuable

523 sentinels of change, particularly when monitored in a quantitative, multi-proxy way (Moatar *et al*

524 2020, Frei et al 2020, Helton et al 2011, Pinay et al 2018, Brookshire et al 2009, Cole et al 2007,

525 Shogren *et al* 2021, Casas-Ruiz *et al* 2023, 2017).

4.3 Context-dependence of biogeochemical proxies in surface and subsurface environments 526 Biogeochemical proxies and tracers are indispensable tools to evaluate hydrological and 527 ecological processes across scales (Pinay et al 2015, Bernardie et al 2018, Frei et al 2020). 528 Particularly when working at the watershed scale or when investigating groundwater 529 environments where extensive physical exploration is impossible, proxies can reveal processes 530 that are otherwise unobservable, including microbial metabolism, hydrological residence time, 531 and weathering processes (Marçais et al 2018, Kolbe et al 2019, Leibundgut et al 2009). Because 532 proxies such as solutes, isotopes, and physicochemical attributes (e.g., temperature, pH, etc.) can 533 reveal mass flux and processing, they have been used extensively to investigate the sources of 534 535 nutrient pollution and to compare reactivity in different components of a watershed (Sebilo et al 2013, Covino 2017, Thomas and Abbott 2018). 536

However, one of the fundamental assumptions about interpreting a tracer is that it will retain or conserve a portion of the signal imbued by the environment of interest as it travels to the location where it can be sampled (Abbott *et al* 2016, Aravena and Robertson 1998). If the tracer is completely consumed or modified, it can communicate no information about its source and the reactions it experienced during transport—unless, of course the product of its consumption can be measured, such as excess N₂ in this study (Kolbe *et al* 2019). Specifically, the limitations of using δ^{15} N as a tracer for identifying nitrogen sources and transformations has

been well documented (Choi *et al* 2017, Robinson 2001). The δ^{15} N signal is often used as an 544 indicator of not only nitrogen cycles but also nitrogen origin. Our results suggest that δ^{34} S is a 545 more effective overall tracer of denitrification than δ^{15} N as it is resilient to the conflicting signals 546 and destructive interference that complicate δ^{15} N. Sulfur and nitrogen cycles are connected in 547 many ways, including through autotrophic denitrification and through redox dependence of both 548 549 denitrification and sulfur reduction (Pu et al 2014, Aquilina et al 2018, Abbott et al 2016). It is therefore not surprising that information contained in δ^{34} S could shed light on multiple aspects of 550 NO₃⁻ transport and degradation. 551

In this study, we observed that even in cases of strong hydrological connectivity (Thomas 552 and Abbott 2018, Vautier et al 2021), the biogeochemical signature of shallow wells, streams, 553 and deep wells remained distinct. This was evidenced in the opposite relationships between $\delta^{34}S$ 554 and NO₃⁻, Cl⁻, and NO₃⁻, as well as the contrasting spatiotemporal variability regimes for many 555 parameters in the surface and subsurface. While we are not surprised that distinct 556 557 biogeochemical reactions and physical conditions dominate surface and subsurface environments (Frey et al 2014, Ben Maamar et al 2015, Aquilina et al 2018, Li et al 2021), we were surprised 558 by the abruptness of the transitions between these environments, even where we know 559 560 hydrological connectivity exists (Kolbe et al 2016, Le Lay et al 2019b). Four non-exclusive hypotheses that could account for this disconnect are: 1. Surface-subsurface interfaces erase or 561 substantially modify biogeochemical signatures (Krause et al 2014), 2. Biogeochemical 562 signatures persist across interfaces but are blurred by contrasting conditions in the new 563 environment (Helton et al 2011), 3. The mass balance of surface-subsurface exchanges is 564 insufficient to substantially influence the conditions in either; i.e., inputs are diluted (Pinay et al 565 2018, Le Moal *et al* 2019), and 4. The groundwater contributing to the streams is different than 566

the groundwater measured in wells dispersed across the aquifer; i.e., streams are deriving their 567 water and solutes from near-stream rather than catchment-wide sources (Ågren et al 2010, Gu et 568 al 2017). 569

570 Regardless of the cause, the contrasting chemistry of the surface and subsurface represents an ecological paradox and practical challenge. On the ecological side, these 571 572 observations may suggest that surface and subsurface environments are largely disconnected and compartmentalized. Even in these wet watersheds in western France with relatively high 573 hydrological connectivity, the interaction between groundwater and stream water chemistry is 574 complex (Kolbe et al 2016, Le Lay et al 2019a). On the practical side, these observations 575 demonstrate that proxy reactivity and hence information content are extremely context dependent 576 (Abbott et al 2016). This was, predictably the situation for the dissolved gases, which quickly 577 equilibrate with the atmosphere after emerging to the surface (Ayraud et al 2008, Vautier et al 578 2021). However, we did not expect the same to hold for nutrients such as NO_3^- , DSi, and SO_4^{2-} 579 580 (Lajtha 2019, Lovett et al 2018, Heiner et al 2022). Together, these observations highlight how biogeochemical and physical processes can modify ecological signals on short timescales and 581 medium spatial scales in surface environments and long timescales and small spatial scales in 582 583 subsurface environments.

584

4.4 Was it my neighbor or my grandparents?

Understanding where nutrients originate and how they are transported through watersheds 585 has ecological and interpersonal implications. Debates about who is responsible for water 586 pollution in the surface and subsurface are likely as old as agriculture and urbanization (Snow 587 1856). This is certainly the case in Europe, where local, national, and E.U. targets and 588 regulations can have real environmental and legal consequences (Musacchio et al 2020, Ebeling 589

et al 2021). Indeed, this issue was raised at a meeting with a farmer from this LTSER when he
realized the importance of legacy nutrient loading in determining current-day water quality.
During a landowner-researcher dinner, he exclaimed something along the lines of, "So you're
telling me I have high nitrates because of my grandparents, not my neighbors?"

The integrated surface and subsurface sampling from our study and the Zone Atelier 594 595 Armorique LTSER more broadly provides some perspective on these questions of nutrient legacies and pollutant transport (Thomas et al 2019). Because of its low residence time and high 596 597 lateral connectivity, the stream network acts much more like an equitable commons, where pollutant sources and sinks average out to create consistent conditions. Conversely, the 598 subsurface is decidedly inequitable, with some areas laden with nutrient legacies that will last 599 decades (Vautier et al 2021), and other areas endowed with high nutrient removal capacity that 600 will protect water quality despite high contemporary or historical loading (Aravena and 601 602 Robertson 1998). This highlights the challenge of assessing nutrient vulnerability or even 603 characterizing four-dimensional nutrient sources and sinks (Hartmann et al 2021, Kolbe et al 2016). However, the linked nature of the subsurface and surface also illuminates a commonsense 604 shortcut to reducing eutrophication, which we explore in the following section (Koh et al 2018, 605 606 Basu et al 2022, Pinay et al 2015, Frei et al 2021, Poisvert et al 2017, Sabo et al 2021b).

607

4.5 Living within our nutrient means: focus on reducing nutrient loads

Since the Great Acceleration of the mid-20th century, humanity's capacity to create, collect,
and distribute reactive nutrients has far exceeded the Earth system's ability to retain or remove
them (Steffen *et al* 2015b, Elser and Bennett 2011, Vitousek *et al* 1997, Steffen *et al* 2015a). This
global nutrient overload is at the root of many of the "syndromes" of the Anthropocene (Hale *et al*2016, Meybeck 2003, Foley *et al* 2011). Rather than reducing our overuse, it is tempting to try to

reengineer the environment to tolerate our excesses. However, supercharging nutrient retention 613 and removal has multiple limitations and side effects. For example, while the overapplication of 614 fertilizer is relatively straightforward to monitor using national and regional nutrient inventories 615 (Poisvert et al 2017, Sabo et al 2021a), we do not have reliable methods for predicting nutrient 616 retention capacity at watershed or regional scales (Seitzinger et al 2010, Pinay et al 2015). At these 617 618 scales, there are orders-of-magnitude variations in hydrological residence time and nutrient removal rates that do not correlate with known proxies (Hartmann et al 2021, Burt 2001, Burt and 619 620 Pinay 2005, Thomas et al 2015, Bernhardt et al 2017). The physical, chemical, and biological 621 peculiarities of each watershed and watershed component-especially in the subsurfaceprecludes prediction of nutrient resilience or vulnerability (Pinay et al 2015, Bernardie et al 2018, 622 Frei et al 2020, Wolters et al 2022). Even if we were able to measure nutrient retention reliably at 623 a moment in time, this would not necessarily allow us to set sustainable nutrient loading thresholds 624 because nutrient release depends on contemporary and historical conditions. For even a small 625 626 watershed, nutrient transit times are often in the decades or centuries, creating a situation that is ecologically and politically untenable because of the lag between policy changes and ecosystem 627 response (Sebilo et al 2013, Basu et al 2022, Van Meter et al 2021, Nguyen et al 2022b, Ascott et 628 629 al 2021). Seeking to supercharge removal by augmenting electron donors or modifying hydrology entails risks and tradeoffs as well (Pu et al 2014, Roley et al 2016, Aravena and Robertson 1998). 630 631 Finally, relying on microbial processes to remove excess NO_3^- can exacerbate climate change if 632 the NO₃⁻ is partially denitrified into the greenhouse gas N_2O and not to the inert N_2 (Hallberg *et al* 2022, Lu et al 2022, Gerber et al 2016). 633

634 These complexities indicate that relying on nutrient retention and removal to resolve635 nutrient pollution is a partial solution at best. Protecting natural zone of denitrification (e.g.,

groundwater aquifers and riparian zones) is highly desirable because of the multiple cobenefits 636 (Pinay et al 2018, Cheng et al 2020), but it is clear that our capacity to load ecosystems with 637 reactive nutrients far exceeds their ability to remove them. Reducing NO₃⁻ input into the soil and 638 water is the most prudent action to reduce eutrophication (Wan et al 2022). This can and should 639 be done by implementing best management practices, and potentially limiting fertilizer 640 641 application to cultivation of human food rather than feed for animal agriculture, decorative plants, and biofuels (Frei et al 2021, 2020). In this case, the proverb holds: "An ounce of 642 prevention is worth a pound of cure." 643

644

645 Acknowledgements

646 This project was funded by the European Union's Seventh Framework Program for research,

technological development and demonstration under grant agreement no. 607150 (FP7-PEOPLE-

648 2013-ITN – INTERFACES - Ecohydrological interfaces as critical hotspots for transformations

of ecosystem exchange fluxes and biogeochemical cycling). Abbott, Severe, Errigo, and Proteau,

were supported by the U.S. National Science Foundation (EAR 2011439 and EAR 2012123).

- 651 Pinay was also supported by the French EC2CO grant "Caractérisation hydrologique et
- biogéochimique de la dénitrification dans les paysages." We thank P. Petitjean for data
- 653 management.

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