

# Evaluation of the performance of different atmospheric chemical transport models and inter-comparison of nitrogen and sulphur deposition estimates for the UK

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## Abstract

An evaluation has been made of a number of contrasting atmospheric chemical transport models, of varying complexity, applied to estimate sulphur and nitrogen deposition in the UK. The models were evaluated by comparison with annually averaged measurements of gas, aerosol and precipitation concentrations from the national monitoring networks. The models were evaluated in relation to performance criteria. They were generally able to satisfy a criterion of 'fitness for purpose' that at least 50% of modelled concentrations should be within a factor of two of measured values. The second criterion, that the magnitude of the normalised mean bias should be less than 20%, was not always satisfied. Considering known uncertainties in measurement techniques, this criterion may be too strict. Overall, simpler models were able to give a good representation of measured gas concentrations whilst the use of dynamic meteorology, and complex photo-chemical reactions resulted in a generally better representation of measured aerosol and precipitation concentrations by more complex models.

The models were compared graphically by plotting maps and cross-country transects of wet and dry deposition as well as calculating budgets of total wet and dry deposition to the UK for sulphur, oxidised nitrogen and reduced nitrogen. The total deposition to the UK varied by +/- 22-36% amongst the different models depending on the deposition component. At a local scale estimates of both dry and wet deposition for individual 5km x 5 km model grid squares were found to vary between the different models by up to a factor of 4.

## 1. Introduction

Concern over the emissions of pollutant gases leading to acidification of soils and surface waters in Europe arose during the 1970s and 1980s, principally due to SO<sub>2</sub> emissions from commercial power production caused by burning coal. The environmental degradation of sensitive ecosystems,

56 notably in upland regions was linked to the emissions of pollutants which in some cases originated  
57 in neighbouring countries, at distances of up to approximately a thousand km away.

58 Following substantial reductions in SO<sub>2</sub> emissions (<http://naei.defra.gov.uk/>) scientific  
59 interest has subsequently become more focused on eutrophication of natural ecosystems due to the  
60 deposition of nitrogen (N) from both oxidized nitrogen (emitted primarily as NO<sub>x</sub> from fuel  
61 combustion) and reduced nitrogen (emitted mostly as NH<sub>3</sub> from agricultural sources). Heath-land  
62 communities are highly sensitive to N deposition. Field experiments have correlated inorganic  
63 nitrogen deposition to a loss of biodiversity in different ecosystem ranging from grassland (Stevens  
64 *et al.*, 2004) to boreal forest (Nordin *et al.*, 2005). Nitrogen deposition is also an important pathway  
65 leading to acidification of terrestrial and freshwater ecosystems. The eutrophication of fresh waters  
66 can cause a severe reduction in water quality impacting on fish stocks and other plant and animal  
67 life. Atmospheric deposition of reactive nitrogen has been recognised as one of the most significant  
68 threats to global biodiversity (Sala *et al.*, 2000).

69 Deposition of sulphur and nitrogen to the earth's surface can occur via the mechanisms of  
70 both 'dry' deposition and 'wet' deposition. Dry deposition is primarily due to gaseous compounds  
71 (SO<sub>2</sub>, HNO<sub>3</sub>, NH<sub>3</sub>, and NO<sub>2</sub>) with aerosol making smaller contributions. In the case of NH<sub>3</sub>, the  
72 deposition is largest near to emissions sources which occur in the rural environment (e.g. Loubet *et*  
73 *al.* 2009, Vogt *et al.*, 2013). Oxidized nitrogen (NO<sub>x</sub>) emissions are primarily in the form of NO  
74 which has a very low deposition velocity to vegetation so atmospheric oxidation to HNO<sub>3</sub> must take  
75 place before significant deposition occurs. For SO<sub>2</sub>, emissions are predominantly due to power  
76 generation from elevated point sources with the speed of vertical diffusion to the surface being  
77 strongly influenced by meteorological conditions. Wet deposition occurs due to the incorporation of  
78 aerosol particles (acting as cloud condensation nuclei and scavenged below cloud) which fall to  
79 ground as precipitation, as well as below-cloud and in-cloud scavenging of soluble gases.

80 National and international monitoring networks have been set up during the last few  
81 decades to analyse the chemical composition of precipitation, notably sulphate (SO<sub>4</sub><sup>2-</sup>), nitrate (NO<sub>3</sub><sup>-</sup>)  
82 and ammonium (NH<sub>4</sub><sup>+</sup>). In the UK, the network in its current configuration was initiated in 1986  
83 and is now a component network of the UK Government's Eutrophying and Acidifying  
84 Atmospheric Pollutants (UKEAP) project (<http://uk-air.defra.gov.uk/networks/network-info?view=ukeep>, last access 5/3/2015). Monitoring of gas phase pollutants initially focused on SO<sub>2</sub>  
85 however since then networks to monitor NO<sub>2</sub>, NH<sub>3</sub> and other pollutant concentrations at rural  
86 locations have been set up. The DELTA system (Sutton *et al.*, 2001) was developed initially to  
87 monitor NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> in regional, long-term monitoring being subsequently extended to sample  
88 acid gases (SO<sub>2</sub>, HNO<sub>3</sub>, HCl), and aerosols as well as the inorganic components of aerosol (size

90 fraction < 4  $\mu\text{g m}^{-3}$ ). For the period of the model inter-comparison addressed here (2003, see below)  
91 the DELTA method was applied for all these air pollution components at 12 sites, with the network  
92 since having been extended to 30 sites.

93 International legislation has been successful in reducing emissions of  $\text{SO}_2$  to the atmosphere  
94 through the United Nations Economic Commission for Europe Gothenburg Protocol (1999) and the  
95 European Union National Emissions Ceiling Directive (2001). In the UK, an 85% reduction in  $\text{SO}_2$   
96 emissions occurred between 1970 and 2003 primarily due to fuel switching from coal to gas and the  
97 introduction of flue gas desulphurization to power generating plants. The reduction in emissions has  
98 led to major decreases of sulphur concentrations measured in the atmosphere in both air and  
99 precipitation (RoTAP, 2012), with corresponding reductions in acidifying inputs to natural  
100 ecosystems in the UK and other European countries. Major reductions in emissions of  $\text{NO}_x$  of 40%  
101 between 1970 and 2003 have also occurred due to introduction of more efficient combustion  
102 processes and the fitting of catalytic convertors on vehicles. However these reductions have not  
103 resulted in major decreases in wet deposition of oxidized nitrogen, most likely because of non-  
104 linearity in atmospheric chemical reactions, in particular the interactions between gas phase and  
105 aerosol lifetimes (Fowler *et al.*, 2005). Furthermore, decreases in estimated emissions of ammonia  
106 in the UK have been more modest (11% between 1990 and 2010) and reductions in concentrations  
107 of ammonia in air and wet deposition of reduced nitrogen have not been observed on a national  
108 scale. As a result, the decrease of inputs of nitrogen to natural ecosystems has been much less  
109 significant than that for sulphur deposition during recent decades (Matejko *et al.*, 2009; Fowler *et*  
110 *al.*, 2005). Analysis of data from the EMEP (European Monitoring and Evaluation Programme)  
111 monitoring network has shown that whilst ammonium and nitrate concentrations in precipitation  
112 have declined in Europe, the sum of nitrate and nitric acid in air remained at the same level (Fagerli  
113 and Aas, 2008).

114 Atmospheric Chemical Transport Models (ACTMs) are computer programs which have  
115 been developed to simulate meteorological, physical and chemical processes. They are able to  
116 provide estimates of the concentration and deposition of air pollutants known to have detrimental  
117 impacts on both human health and natural ecosystems. In this study a range of simpler and more  
118 complex ACTMs have been applied to make estimates of sulphur and nitrogen deposition. An  
119 operational evaluation of the performance of the models has been undertaken by comparison with  
120 measurements of concentrations in air (gas and aerosol) and precipitation. A comparison of wet and  
121 dry deposition obtained with the different models has been made using both national deposition  
122 totals and a cross-country transect.

123           **2. Description of Models**

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125           ACTMs have been used in the UK during the last two decades to calculate acid deposition  
 126 and provide advice to policy makers. The advantages of models include:

- 127           (i)      Estimation of the concentration and deposition of air pollutants at a large number of  
 128 model grid cells in the UK (typically ~ 10,000 for a model with a 5 km grid  
 129 resolution).
- 130           (ii)     Estimation of the future changes of impacts on ecosystems based on projections for  
 131 pollutant emissions.
- 132           (iii)    Attribution of pollutant deposition to individual emissions sources through source  
 133 emission reduction simulations.

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135           In contrast, monitoring of air pollutants is both spatially and temporally limited by the number of  
 136 sites and the period of their operation.

137           Atmospheric chemical transport modelling in the UK was initially undertaken using  
 138 ‘simpler’ models such as HARM (Metcalfe *et al.*, 2001) and FRAME (Singles *et al.*, 1998). These  
 139 Lagrangian models use straight line trajectories and operate in an annual average mode, assuming  
 140 constant drizzle to drive wet deposition (based on maps of precipitation for the UK) and annual  
 141 wind direction frequency roses (Dore *et al.*, 2006) to represent general circulation patterns of air  
 142 trajectories. These models simulate a ‘moving air column’ and independently perform calculations  
 143 along pre-defined trajectories by contrast to Eulerian models which simultaneously perform  
 144 calculations at all points in a predefined grid.

145           Major advances in High Performance Computer technology as well as a general move to  
 146 open source code for both meteorological models and ACTMs have both driven a move to the use  
 147 of more complex models during recent years, with an emphasis on Eulerian approaches. These  
 148 models include the US Environmental Protection Agency Community Multi-scale Air Quality  
 149 (CMAQ) modelling system (Byun *et al.*, 2006) and the EMEP model (Simpson *et al.*, 2012),  
 150 including its high resolution application to the UK which is used in the present study (EMEP4UK,  
 151 Vieno *et al.* 2014). Such systems use a meteorological model to generate 3-dimensional temporally  
 152 evolving data on wind speed, temperature, humidity, cloud and precipitation which are then used to  
 153 drive the ACTM. The meteorological data was evaluated with the Met Office Integrated Data  
 154 Archive System (MIDAS, <http://catalogue.ceda.ac.uk/uuid/220a65615218d5c9cc9e4785a3234bd0>  
 155 last access 2 July 2015). These ‘more complex’ models also include Lagrangian approaches, such

156 as NAME (Redington *et al.*, 2009) which is driven by temporally evolving meteorology in a  
157 Lagrangian framework. For the present study CMAQ and EMEP4UK used independently  
158 generated meteorological data calculated with the WRF (Weather Research and Forecasting) model  
159 (<http://www.wrf-model.org>; Skamrock and Klemp, 2008) while the NAME model was run using  
160 global meteorological data calculated with the UK Met Office Unified Model.

161 The use of both simpler and more complex models provides complementary benefits. For  
162 example the advantages of more complex models include: a more detailed representation of  
163 meteorology and its influence on concentrations of air pollutants; high temporal resolution of  
164 pollutant concentration (Vieno *et al.*, 2014); more detailed parameterisation of non-linear  
165 atmospheric chemical reactions; simultaneous multi-pollutant simulation (i.e. representation of acid  
166 deposition, surface ozone and particulate matter in one model; Byun *et al.*, 2006). In contrast, the  
167 simpler models benefit from a fast simulation time which allows: multiple simulation applications  
168 including source-receptor and integrated assessment studies (i.e. Oxley *et al.*, 2013); uncertainty  
169 studies (Page *et al.*, 2008); high spatial resolution studies and detailed vertical resolution  
170 (Hallsworth *et al.*, 2010; Dore *et al.*, 2012).

171 The models involved in this study included two simpler Lagrangian models (employing  
172 annually averaged meteorology) and three more complex models driven by dynamic meteorology  
173 and using diurnally variable photo-chemical reaction schemes. A summary of the models is given  
174 in Table 1. This inter-comparison included two independent applications of the CMAQ model and  
175 also two applications of the EMEP model run at different resolutions of 50 km and 5 km  
176 respectively (termed EMEP.MSCwest and EMEP4UK). The EMEP.MSCwest simulation used data  
177 from the HILRAM meteorological model (<http://www.hirlam.org/> last access 2 July 2015). This  
178 allowed an assessment of the sensitivity of model grid resolution and of the variability in modelled  
179 concentrations and deposition not just between different models but for different applications of the  
180 same model. The two CMAQ simulations used identical meteorological inputs but different annual  
181 emissions profiles (discussed below). The models used common inputs of annual atmospheric  
182 emissions from the UK National Atmospheric Emissions Inventory (<http://naei.defra.gov.uk/> last  
183 access 5: March 2015) which are updated annually and gridded at a 1 km resolution. The models  
184 were unconstrained with regard to choice of boundary conditions, meteorological data, land use  
185 cover and internal model parameters. The model domains covered the entire United Kingdom  
186 (including the northern islands) for HARM and the British Isles (including the Republic of Ireland)  
187 for CMAQ, EMEP4UK, FRAME and NAME. The model domains were not uniform but typically  
188 covered an area of approximately 900 km west-east x 1200 km south-north. The models included in

this study are of varying levels of chemical complexity and therefore have differences regarding the speciation of the chemical components of the atmosphere. To generate the boundary conditions for a UK scale simulation, the ACTMS were first run at a coarser 50 resolution over a European domain with the EMEP4UK and CMAQ European simulations using meteorological data from a 50 resolution WRF simulation. All the models used in this study include the major inorganic sulphur and nitrogen compounds. These include gases which are significant for dry deposition ( $\text{SO}_2$ ,  $\text{NO}_2$ ,  $\text{HNO}_3$  and  $\text{NH}_3$ ), as well as particulate matter components which are efficiently wet deposited ( $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ ) and can also be dry deposited. Other chemical components including many  $\text{NO}_y$  species such as nitrous acid ( $\text{HONO}$ ) are included in the more complex models but in this exercise their dry deposition was not explicitly modelled. The simpler models adopt a single scavenging parameter for wet deposition processes whilst the more complex models have separate scavenging coefficients for in cloud and below cloud scavenging of gases and particles. Various different resistance formulae are used to calculate dry deposition velocities. However whilst the more complex models use temporally evolving meteorology in their calculations, dry deposition calculated with simpler models is based on annually averaged deposition velocities.

Other models used to calculate sulphur and nitrogen deposition include the Danish Ammonia Modelling System (DAMOS, Geels *et al.*, 2012) which combined a long range transport model with a local scale Gaussian model for dry deposition. Kranenburg *et al.* (2013) describe the development of a source apportionment tool in the LOTOS-EUROS model which was used to track the emissions sources contributing to nitrogen concentrations in the Netherlands. The CHIMERE model was used by Garcia-Gomez *et al.* (2014) to assess the threat of nitrogen deposition to the Natura 2000 network of nature reserves in Spain. Appel *et al.* (2010) assessed the performance of CMAQ over the USA by comparison with measurements of wet deposition of sulphur and nitrogen from the National Atmospheric Deposition Programme.

**Table 1:** Summary of models participating in the inter-comparison including model grid resolution. Two independent applications of the CMAQ and EMEP models (the latter at different grid resolutions) are included.

Model name & grid	Classification	Type	Reference
CMAQ.JEP (5 km)	More Complex	Eulerian dynamic meteorology	Chemel <i>et al.</i> (2010)
CMAQ.UH (5 km)	More Complex	Eulerian with dynamic meteorology	Chemel <i>et al.</i> (2010)
EMEP.MSCwest (50 km)	More Complex	Eulerian dynamic meteorology	Simpson <i>et al.</i> (2012)
EMEP4UK (5 km)	More Complex	Eulerian dynamic meteorology	Vieno <i>et al.</i> (2014)
FRAME (5 km)	Simpler	Lagrangian statistical meteorology	Matejko <i>et al.</i> (2009)
HARM (10 km)	Simpler	Lagrangian statistical meteorology	Page <i>et al.</i> (2008)
NAME (5 km)	More Complex	Lagrangian dynamic meteorology	Redington <i>et al.</i> (2009)

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### 216     **3. Measurement Data**

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Atmospheric monitoring data for 2003 were used in this study for comparison with the model estimates. Measured concentrations in air and precipitation were obtained as part of the component networks which now collectively comprise UKEAP:

- $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ ,  $\text{NH}_4^+$  precipitation concentrations from bulk sampler analysis at 37 sites.

- aerosol ( $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ ,  $\text{NH}_4^+$ ) at 12 sites using DELTA samplers

- $\text{SO}_2$ ,  $\text{NH}_3$  and  $\text{HNO}_3$  gas concentrations at 12 sites using DELTA samplers

- $\text{SO}_2$  gas concentrations at 37 sites using bubbler samplers

- $\text{NH}_3$  gas concentrations at 88 sites using both active (DELTA) samplers and passive (ALPHA) samplers

- $\text{NO}_2$  gas concentrations at 32 sites using diffusion tubes

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Further details of the monitoring networks are included in Hayman *et al.* (2004), which is available from <http://uk-air.defra.gov.uk/networks/network-info?view=ukeap> (last access: 5/3/2015). Data capture averaged across the sites exceeded 97% of the samples collected for particulate and gaseous chemical concentrations. For precipitation chemistry data capture (on average 78%) was lower principally due to exclusion of samples with high phosphate concentrations indicating contamination from bird strike. All monitoring sites used in this study are based at rural or semi-rural locations which are located at least 2.5 km away from significant emissions sources, such as major roads. The location of the monitoring sites are shown in the supplementary material (Figures 1(a)-(f)). Bulk samplers used to measure precipitation composition are sampled fortnightly, DELTA and ALPHA samplers record monthly averages and  $\text{NO}_2$  diffusion tubes are changed every 4-5 weeks.

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251        The evaluation was undertaken with the Openair software using the R statistical language  
 252 (Carslaw and Ropkins, 2012). A report blending text and data analysis was automatically generated  
 253 (Xie, 2013). This approach has the advantage that the results are easily reproducible by a third party  
 254 and updates to submitted model data can rapidly be incorporated by re-running the software.  
 255 Development of the Openair software and its application to inter-comparison of the models in this  
 256 study as well as models for surface ozone and local dispersion is discussed in detail in Carslaw *et*  
 257 *al.* (2011). The more complex models participating in this study generate data with high temporal  
 258 frequency (typically with resolution of a few hours) whilst the simple models are designed to  
 259 calculate only annually averaged concentrations and deposition. For this study, the models have  
 260 been evaluated using only annually averaged data for a single year. 2003 was selected based on the  
 261 availability of meteorological data to drive the complex model simulations.

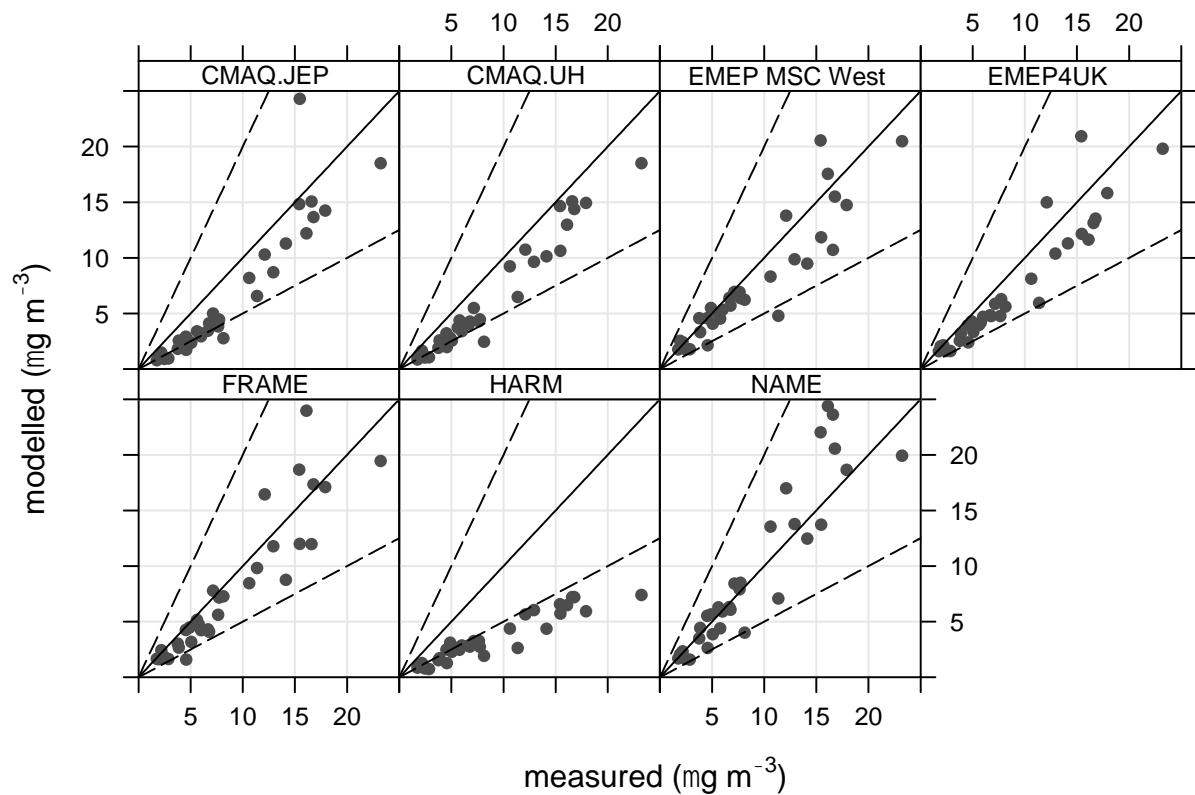
262        A variety of different metrics have been proposed to evaluate the performance of  
 263 atmospheric chemical transport models by comparing the difference between model predictions and  
 264 observations (i.e. Chang and Hanna, 2004). Here we adopt relatively simple criteria for a model to  
 265 be considered ‘fit for purpose’ which were set according to a previously agreed model evaluation  
 266 protocol (Derwent *et al.*, 2010). These were:  $\text{FAC2} > 0.5$  and  $-0.2 < \text{NMB} < 0.2$  where:  $\text{FAC2}$  (i.e.  
 267 ‘factor of 2’) is the fraction of points greater than 0.5 times and less than 2 times the measured  
 268 value and  $\text{NMB}$  is the Normalised Mean Bias defined as:

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$$\text{NMB} = \frac{\sum_{i=1}^n M_i - O_i}{\sum_{i=1}^n O_i}$$

270        where  $O_i$  represents the  $i_{th}$  observed value and  $M_i$  represents the  $i_{th}$  modelled value for a total of  $n$   
 271 observations. The NMB illustrates model over- or under-estimate relative to measurements and is  
 272 useful for comparing pollutants that cover different concentration scales as the mean bias is  
 273 normalised by dividing by the observed concentration.

274        Example plots of the correlation of the models with a gas concentration ( $\text{NO}_2$ ), a particulate  
 275 concentration ( $\text{SO}_4^{2-}$ ) and a concentration in precipitation ( $\text{NH}_4^+$ ) are illustrated in Figures 1(a)-(c)  
 276 with performance statistics summarised in Table 2. The results for other chemical components of  
 277 gas, aerosol and precipitation concentrations are illustrated in the supplementary material (Figure  
 278 S3, Table S1). Table 3 illustrates correlation statistics for all measured chemical components  
 279 averaged across the different models. Further details of the analysis are available at: [http://uk-air.defra.gov.uk/library/reports?report\\_id=652](http://uk-air.defra.gov.uk/library/reports?report_id=652) (last access 30/6/2015).



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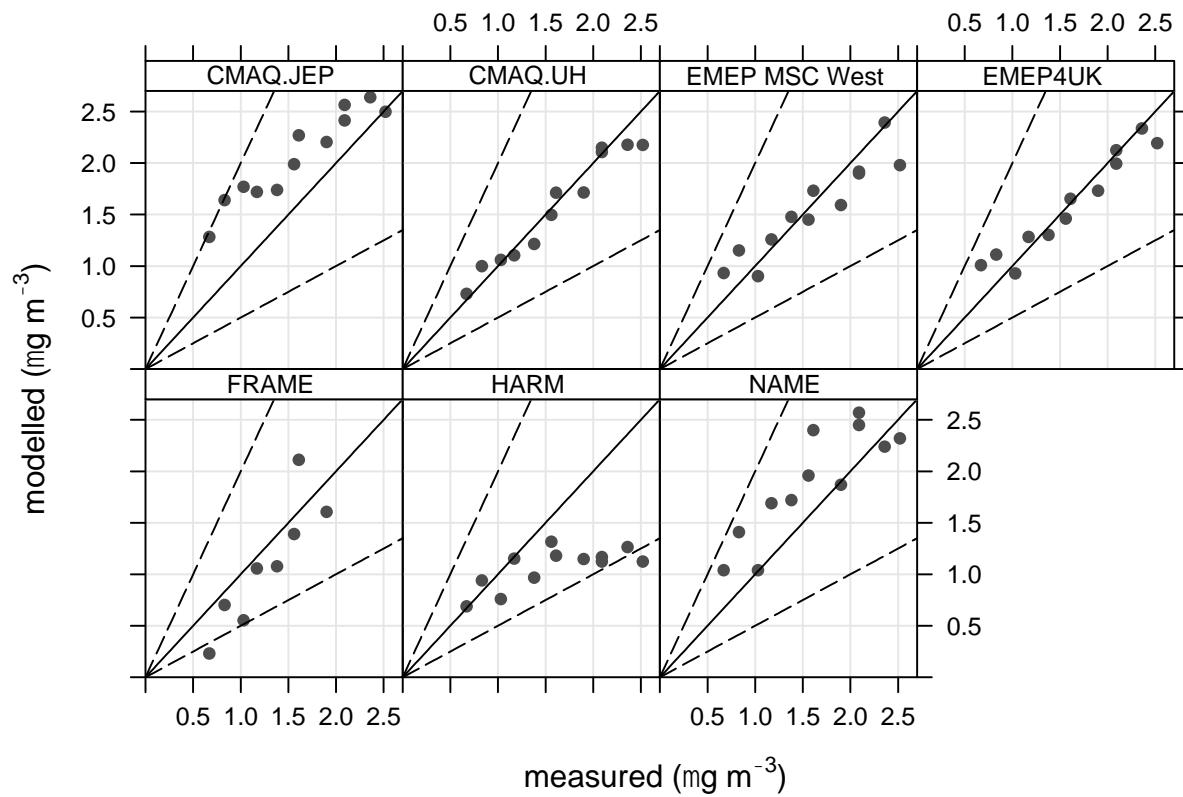
282 **Figure 1(a):** Scatter plot of the annual average modelled concentrations of NO<sub>2</sub> with measurements from the UKEAP  
283 monitoring network.

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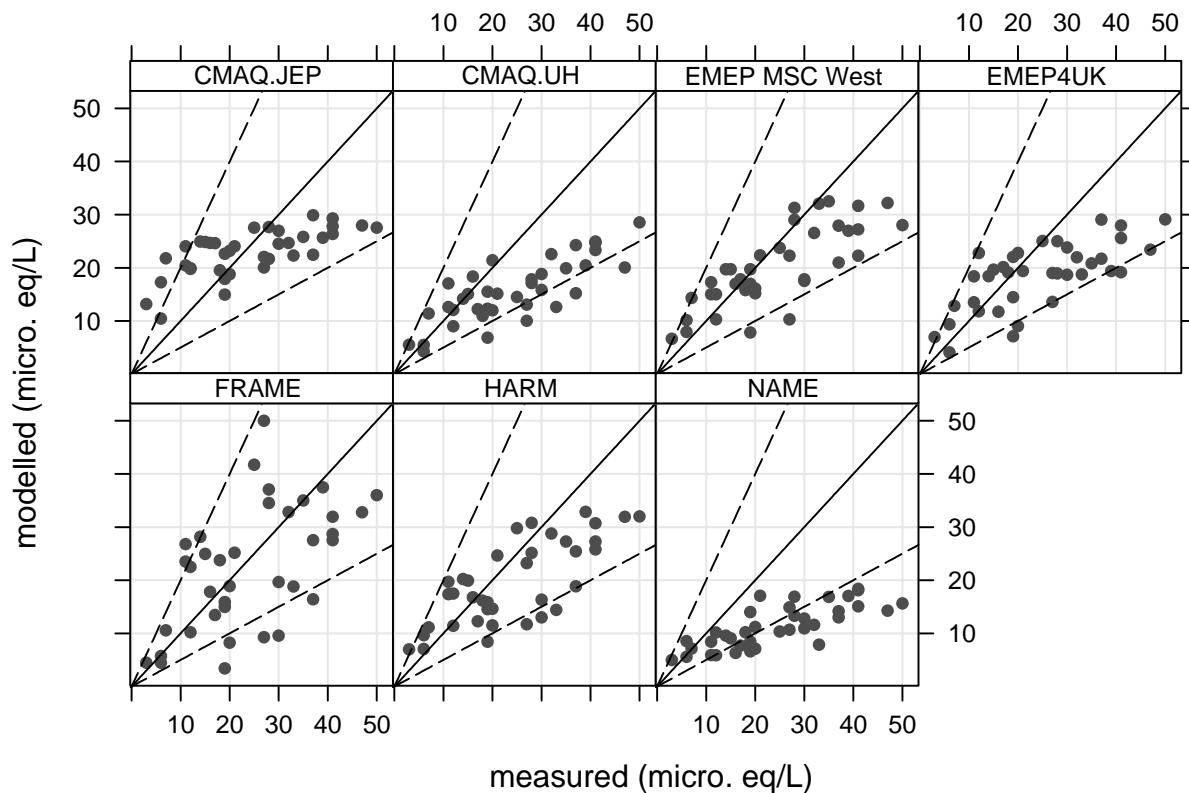
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**Figure 1(b):** Scatter plot of the annual average modelled concentrations of  $\text{SO}_4^{2-}$  aerosol with measurements from the UKEAP monitoring network.



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293 **Figure 1(c):** Scatter plot of the annual average modelled concentrations of  $\text{NH}_4^+$  in precipitation with measurements  
294 from the UKEAP precipitation chemistry monitoring network.  
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298 **Table 2:** Model performance statistics for comparison with concentration measurements: **FAC2**: fraction of points  
299 greater than 0.5x and less than 2x the measured value; **NMB**: normalised mean bias; **r**: Pearson correlation coefficient.

Group	Annual average concentrations of $\text{NO}_2$			Annual average concentrations of $\text{SO}_4^{2-}$ aerosol			Annual average concentrations of $\text{NH}_4^+$ in precipitation		
	FAC2	NMB	r	FAC2	NMB	r	FAC2	NMB	r
CMAQ.JEP	0.72	-0.25	0.92	1.00	0.29	0.95	0.89	-0.05	0.70
CMAQ.UH	0.84	-0.27	0.97	1.00	-0.03	0.97	0.84	-0.36	0.79
EMEP4UK	1.00	-0.18	0.94	1.00	0.00	0.97	0.84	-0.23	0.70
EMEP.MSCwest	0.94	-0.13	0.92	1.00	-0.03	0.93	0.89	-0.17	0.79
FRAME	0.97	-0.10	0.92	0.92	0.14	0.92	0.78	-0.07	0.57
HARM	0.12	-0.60	0.94	0.92	-0.33	0.71	0.86	-0.19	0.75
NAME	0.94	0.0	0.91	1.00	0.20	0.88	0.78	-0.16	0.52

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309 The first of the evaluation criteria,  $\text{FAC2} > 0.5$ , was generally satisfied by the models for all  
 310 variables, but the second condition,  $-0.2 < \text{NMB} < 0.2$ , was not satisfied for all variables (as shown  
 311 in Table 2 and Table S1(a)-(c), supplementary material).

312 The example of comparison with measurements of a gas concentration ( $\text{NO}_2$ , Figure 1(a))  
 313 shows that a simpler model (FRAME) was able to achieve a level of agreement with measurements  
 314 ( $\text{FAC2}=0.97$ ,  $\text{NMB}=-0.1$ ) which is as good as the more complex models. This may be due to the  
 315 fine vertical resolution in FRAME (1 m at the surface) which permits detailed specification of the  
 316 height at which different types of emission source are input to the model. For  $\text{SO}_2$  (supplementary  
 317 material, figure S3.1) there were more significant variations between models for NMB (from 0.07  
 318 to 2.38) and FAC2 (from 0.04 to 0.96).  $\text{SO}_2$  emissions originate primarily from a small number of  
 319 elevated point sources (principally power stations) and treatment of emission height can be  
 320 important. The CMAQ and FRAME models include a plume rise parameterization for point source  
 321 emissions whereas other models apply an emission sector-dependent height. The scatter in  
 322 correlation of the models with  $\text{NH}_3$  gas concentrations (supplementary material, figure S3.2) is  
 323 generally higher ( $\text{FAC2}$  less than 0.78 for all models) than for  $\text{SO}_2$  and  $\text{NO}_2$ . This does not  
 324 necessarily reflect a difficulty in the models to simulate the behaviour of ammonia. It is more likely  
 325 to be caused by the high spatial variability in emissions in rural locations which results in changes  
 326 in ammonia concentrations on scales not captured by ACTMs with grid spacing typically of  
 327 approximately 5 km. The  $\text{NH}_3$  concentration measured at an individual site may not be  
 328 representative of the surrounding area as represented in a 5 km model grid cell (Hallsworth *et al.*  
 329 (2010); Vogt *et al.* (2012)).

330 For aerosol concentrations there is clear evidence that the more complex models obtain  
 331 better correlation with measurements than the simpler models. EMEP4UK, EMEP.MSCwest,  
 332 CMAQ.UH and NAME all achieved a FAC2 of 1.0 and NMBs of 0.0, -0.03, -0.03 and 0.20 for  
 333  $\text{SO}_4^{2-}$  aerosol respectively (figure 1(b)). This may be due to the difficulty of the simple models to  
 334 capture the full magnitude of long range transport of particulate matter from the European continent  
 335 during 2003 when winds from the east were more common than normal. Furthermore complex  
 336 models use hourly meteorological data to drive the formation of secondary inorganic aerosols  
 337 whereas simpler models assume an annual average formation rate. More complex models also  
 338 performed well for  $\text{NO}_3^-$  aerosol (supplementary material, Figure S3.4) notably NAME ( $\text{FAC2}=1.0$ ,  
 339  $\text{NMB} =0.04$ ) and CMAQ.JEP ( $\text{FAC2}=1.0$ ,  $\text{NMB} = -0.20$ ). The overall correlation of the models  
 340 with measurements of  $\text{NO}_3^-$  aerosol (average  $\text{FAC2}=0.75$ ) is not as good as for  $\text{SO}_4^{2-}$  (average  
 341  $\text{FAC2} = 0.94$ ) which may be due to the more complex chemical reactions leading to the formation

342 of oxidised nitrogen aerosol. All models showed some underestimate of  $\text{NH}_4^+$  aerosol  
343 concentrations (average NMB=-0.30, supplementary material Figure S3.3).

344 Considerable scatter in the correlation of all the models for  $\text{NH}_4^+$  concentrations in  
345 precipitation is apparent (Figure 1(c)). None of the models is able to achieve  $\text{FAC2} > 0.9$ . On  
346 average the models tend to underestimate reduced nitrogen concentrations in precipitation as well as  
347 the gaseous and particulate forms which may be an indication that emissions sources are  
348 underestimated or that removal of  $\text{NH}_3$  by dry deposition is too rapid. The average value of  $r$  for all  
349 the models for  $\text{NH}_4$  concentration in precipitation is 0.67, compared to 0.74 and 0.76 for  $\text{SO}_4^{2-}$  and  
350  $\text{NO}_3^-$  respectively (Table 3). All models underestimate  $\text{NO}_3^-$  concentrations in precipitation (average  
351 NMB =-0.26). This may suggest either a missing source of oxidised nitrogen emissions, or overall  
352 underestimates in atmospheric chemical conversion or washout coefficients. The models generally  
353 exhibited negative values of NMB for aqueous phase concentrations (average values of -0.10, -0.26  
354 and -0.18 for  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$  and  $\text{NH}_4^+$  respectively, Table 3). This result may be explained by the fact  
355 that bulk precipitation collectors are used in the monitoring network and will be subject to dry  
356 deposition contamination, principally by gaseous deposition (as discussed below). Overall more  
357 complex models tended to score higher values for  $\text{FAC2}$  and  $r$  than the simpler models for  
358 precipitation concentrations.

359 The influence of model grid resolution can be assessed by comparing the results of the  
360 correlation with measurements for EMEP.MSCwest (50 km grid resolution) and EMEP4UK (5 km  
361 grid resolution). In general, EMEP4UK performed better than EMEP.MSCwest for gas  
362 concentrations ( $\text{SO}_2$ ,  $\text{NO}_2$ , and  $\text{NH}_3$ ) whereas for aerosol and precipitation concentrations, the  
363 differences in correlation are less significant.

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375      **Table 3:** Model performance statistics, averaged for all models, for comparison with measurements: **FAC2**: fraction  
 376      of points greater than 0.5x and less than 2x the measured value; **NMB**: normalised mean bias; **NMGE**: normalised  
 377      mean gross error; **RMSE**: root mean square error; **r**: Pearson correlation coefficient (units for NMGE and RMSE:  $\mu\text{g m}^{-3}$  – gas and aerosol ;  $\mu\text{equiv L}^{-1}$  – aqueous)

Group	Phase	FAC2	NMB	NMGE	RMSE	r
<b>SO<sub>2</sub></b>	Gas	0.63	0.85	0.93	2.25	0.79
<b>NO<sub>2</sub></b>	Gas	0.70	-0.23	0.31	3.42	0.93
<b>NH<sub>3</sub></b>	Gas	0.50	-0.23	0.55	1.50	0.65
<b>SO<sub>4</sub><sup>2-</sup></b>	Aerosol	0.97	0.05	0.23	0.45	0.90
<b>NO<sub>3</sub><sup>-</sup></b>	Aerosol	0.75	-0.21	0.30	1.04	0.94
<b>NH<sub>4</sub><sup>+</sup></b>	Aerosol	0.76	-0.30	0.32	0.52	0.95
<b>SO<sub>4</sub><sup>2-</sup></b>	Aqueous	0.88	-0.10	0.34	10.28	0.74
<b>NO<sub>3</sub><sup>-</sup></b>	Aqueous	0.86	-0.26	0.34	9.43	0.76
<b>NH<sub>4</sub><sup>+</sup></b>	aqueous	0.83	-0.18	0.36	10.69	0.67

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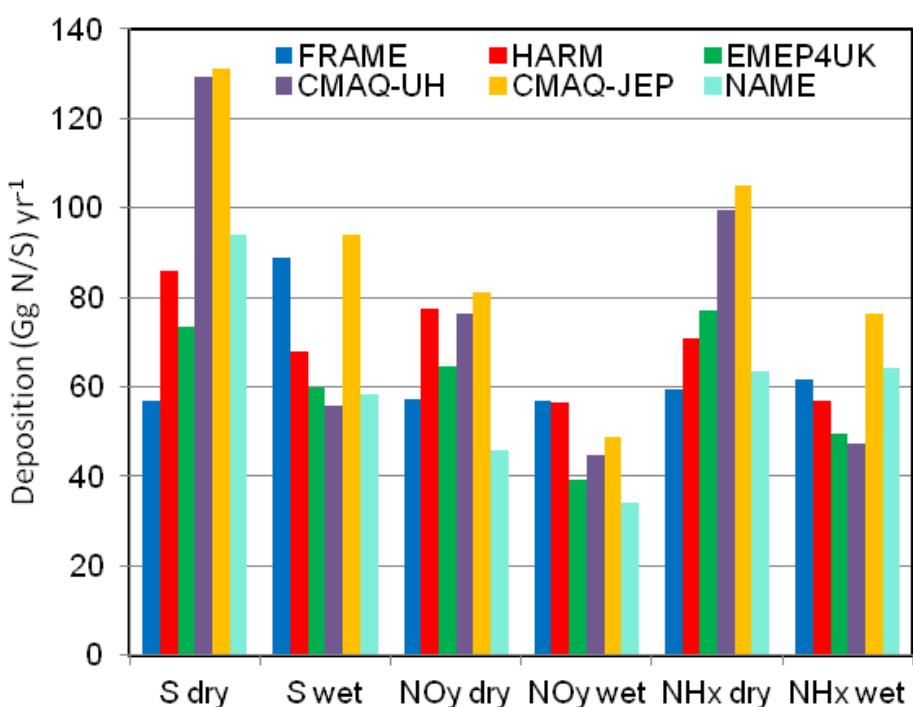
## 4.2 Comparison of modeled deposition

383      The second part of this study involves a comparison of deposition data generated by the models  
 384      described above (with the exception of EMEP.MSCwest) using national scale deposition budgets as  
 385      well as maps and plots along a transect across the UK for wet and dry deposition of SO<sub>x</sub>, NO<sub>y</sub> and  
 386      NH<sub>x</sub> , show in supplementary material, Figure S2. This approach allows visualization of both the  
 387      national scale and local scale variability in deposition between the models.

388      The total UK wet and dry deposition budgets of SO<sub>x</sub>, NO<sub>y</sub> and NH<sub>x</sub> are illustrated in Figure 2  
 389      for all models. The average values and standard deviations for modeled deposition were:  $95 \pm 30$   
 390      Gg S for SO<sub>x</sub> dry deposition and  $71 \pm 17$  Gg S for SO<sub>x</sub> wet deposition ;  $67 \pm 14$  Gg N for NO<sub>y</sub> dry  
 391      deposition and  $47 \pm 9$  Gg N for NO<sub>y</sub> wet deposition ;  $79 \pm 19$  Gg N for NH<sub>x</sub> dry deposition and  
 392       $59 \pm 11$  Gg N for NH<sub>x</sub> wet deposition. These values show that the models predict that dry  
 393      deposition is overall a more important process for removal of sulphur and nitrogen compounds  
 394      from the atmosphere (with higher values than those for wet deposition by 35%, 44% and 34% for  
 395      SO<sub>x</sub>, NO<sub>y</sub> and NH<sub>x</sub> respectively) for the year 2003. However it should be noted that for SO<sub>x</sub> and  
 396      NO<sub>y</sub> dry deposition the higher values occur in industrial and urban areas where as many ecosystems  
 397      sensitive to acid deposition and nitrogen deposition are located in upland areas where wet  
 398      deposition is the most important process. Furthermore the year 2003 was noted for its low annual  
 399      precipitation and this result may not be typical of other years. Differences in dry deposition of SO<sub>2</sub>  
 400      amongst the models occur due to significant variation in modeled surface SO<sub>2</sub> concentrations as is

401 evident from the NMB values for the correlation in measurements (Table S1(a)) and can be  
 402 attributed to different model treatment of elevated point source emissions.

403 The two CMAQ simulations achieved close agreement for dry deposition and NO<sub>y</sub> wet  
 404 deposition. However wet deposition of NH<sub>x</sub> and SO<sub>x</sub> is notably higher with CMAQ.JEP than with  
 405 CMAQ.UH. Whilst the meteorological data used were common to the two models and parameter  
 406 settings for the CMAQ simulations were generally similar, significant differences occurred due to  
 407 the seasonal profile of ammonia emissions. Annual average ammonia emissions were identical but  
 408 the CMAQ.UH simulation had a large seasonal variation in ammonia emissions, with summer time  
 409 emission rates higher by a factor of ten than the winter time values (Figure S4, supplementary  
 410 material). This was effective in restricting the rate of formation of ammonium sulphate aerosol  
 411 during the winter months. For CMAQ.JEP ammonia emission rates during summer months were  
 412 approximately two times higher than winter time values and there was less restriction of ammonium  
 413 sulphate formation during the winter. In reality emissions of ammonia are highly sensitive to  
 414 meteorological conditions, particularly temperature. This issue is discussed in detail by Sutton *et al.*  
 415 (2013). Skjoth (2004) describes a system for dynamically generating seasonally and diurnally  
 416 variable ammonia emissions for use in an ACTM using modelled meteorological data.  
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419 **Figure 2:** The total UK annual wet and dry deposition budgets of SO<sub>x</sub>, NO<sub>y</sub> and NH<sub>x</sub> for the different models (Gg N/S).

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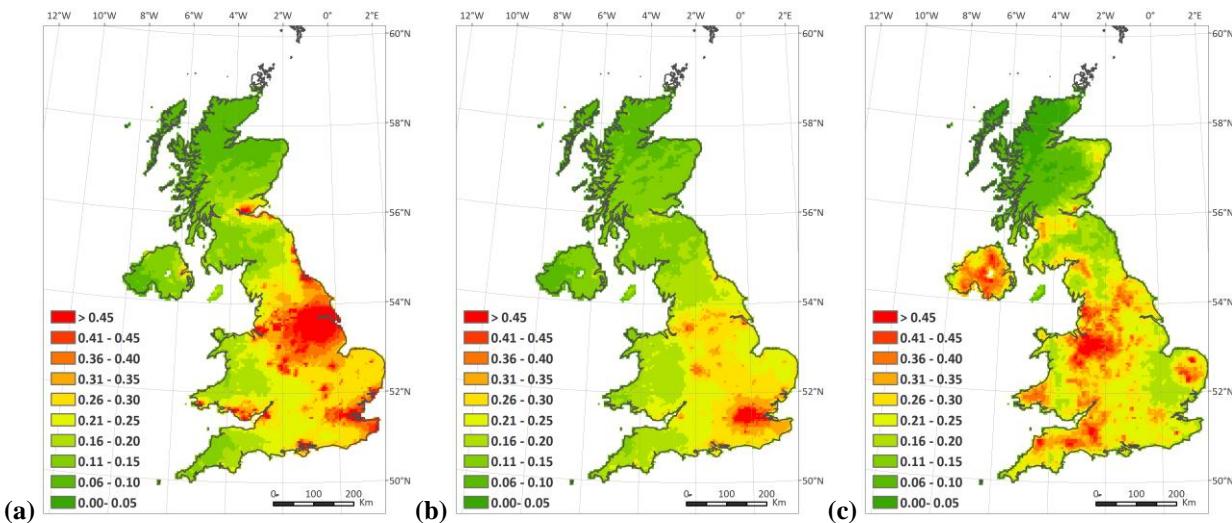
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422 The spatially distributed deposition data for all the models were used to calculate the mean and  
423 standard deviation (figures 3.1 and 3.2) across the UK. Sulphur dry deposition is highest in the  
424 industrial regions of northern England as well as near the coast and at major ports due to the major  
425 contribution to SO<sub>2</sub> emissions from international shipping. NO<sub>y</sub> dry deposition is highest in the  
426 region of major cities and urban areas. For NH<sub>x</sub> dry deposition, the highest values occur in areas of  
427 intensive livestock farming, including Northern Ireland and western and eastern England. The  
428 geographical distribution of wet deposition is similar for sulphur and both oxidised and reduced  
429 nitrogen. Due to the long range transport of aerosol, the highest values occur in the high rainfall  
430 areas of the hills of Wales and northern England. The normalized standard deviation of model  
431 deposition gives an indication of the uncertainty in modeled deposition associated with choice of  
432 model. Sulphur dry deposition and reduced nitrogen deposition show the greatest variability of  
433 deposition in source regions. For wet deposition the highest values of standard deviation amongst  
434 the models occur in the hill regions of Scotland and the far north-west of the country. These  
435 differences are caused both by: variations in formation and long range transport of particulate  
436 matter; differences in representation of atmospheric washout of sulphur and nitrogen compounds by  
437 the models; different estimates of precipitation, particularly orographic precipitation over hill  
438 regions.

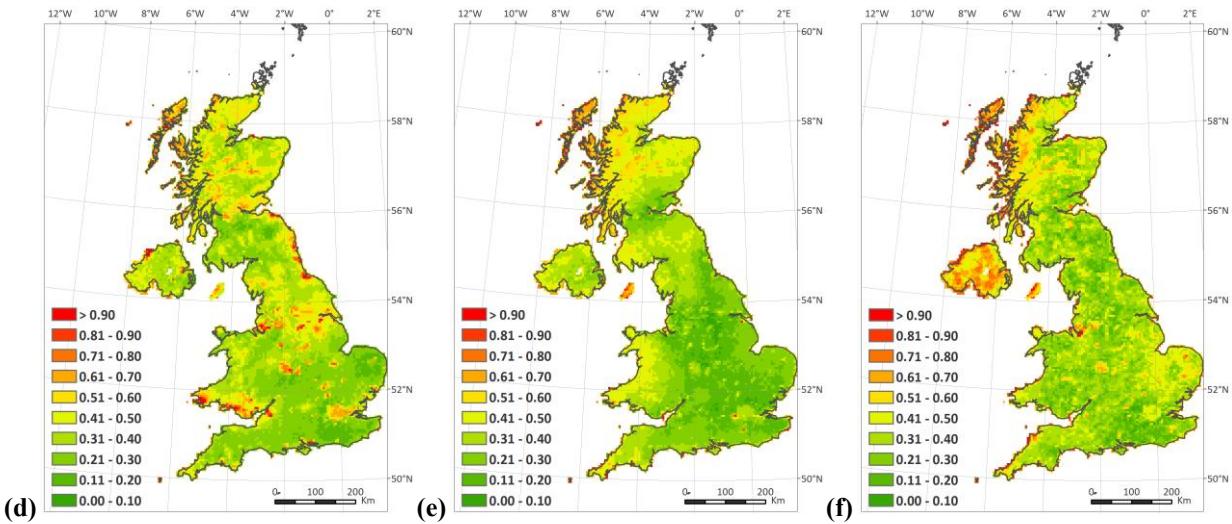
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**Figure 3.1:** Annual dry deposition ( $\text{keq ha}^{-1}$ ) calculated by averaging all the models for: (a)  $\text{SO}_x$ ; (b)  $\text{NO}_y$ ; (c)  $\text{NH}_x$ ; Normalized standard deviation of dry deposition calculated from all models for: (d)  $\text{SO}_x$ ; (e)  $\text{NO}_y$ ; (f)  $\text{NH}_x$

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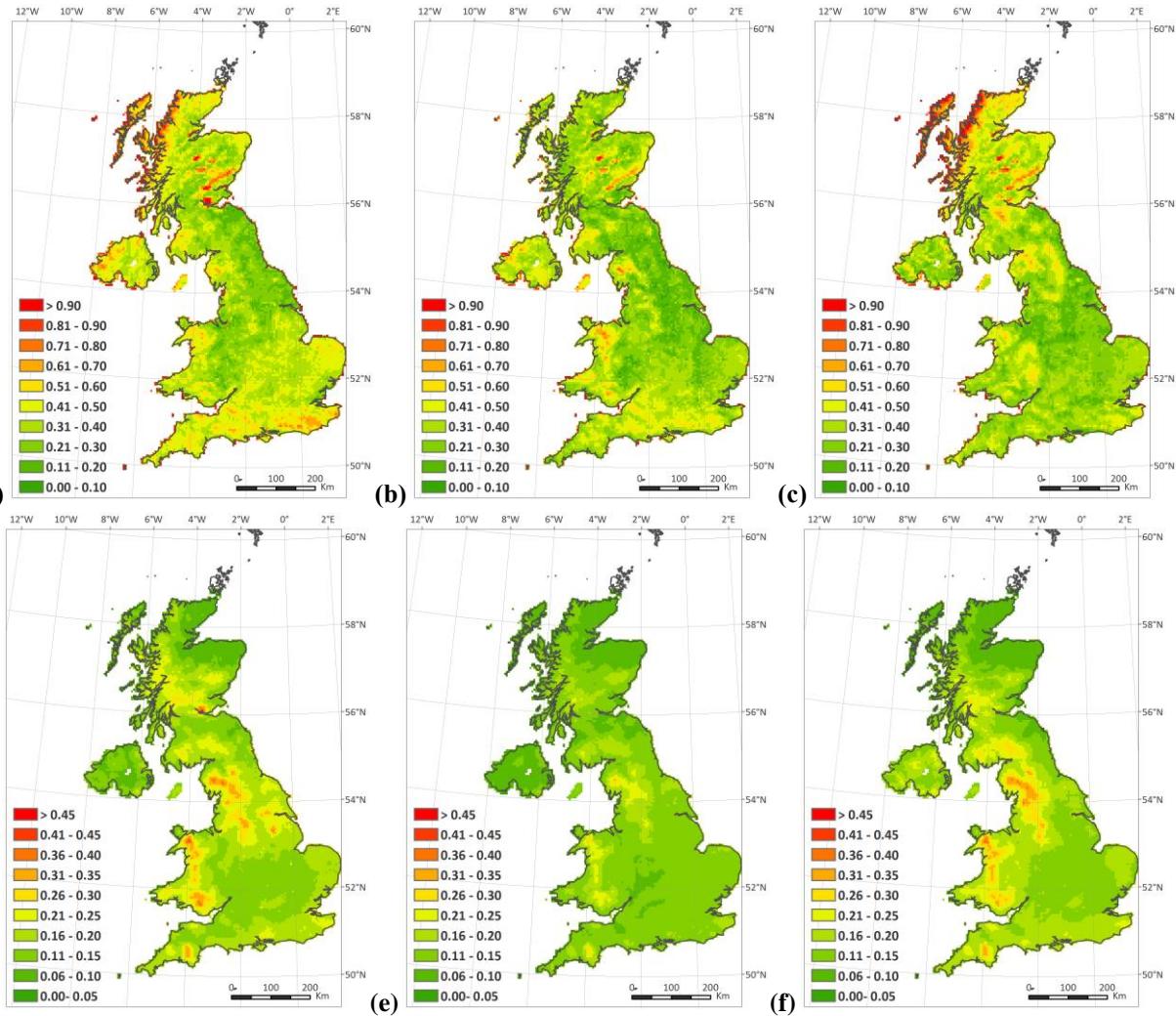
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**Figure 3.2:** Annual wet deposition ( $\text{keq ha}^{-1}$ ) calculated by averaging all the models for: (a)  $\text{SO}_x$ ; (b)  $\text{NO}_y$ ; (c)  $\text{NH}_x$ ; Normalized standard deviation of wet deposition calculated from all models for: (d)  $\text{SO}_x$ ; (e)  $\text{NO}_y$ ; (f)  $\text{NH}_x$

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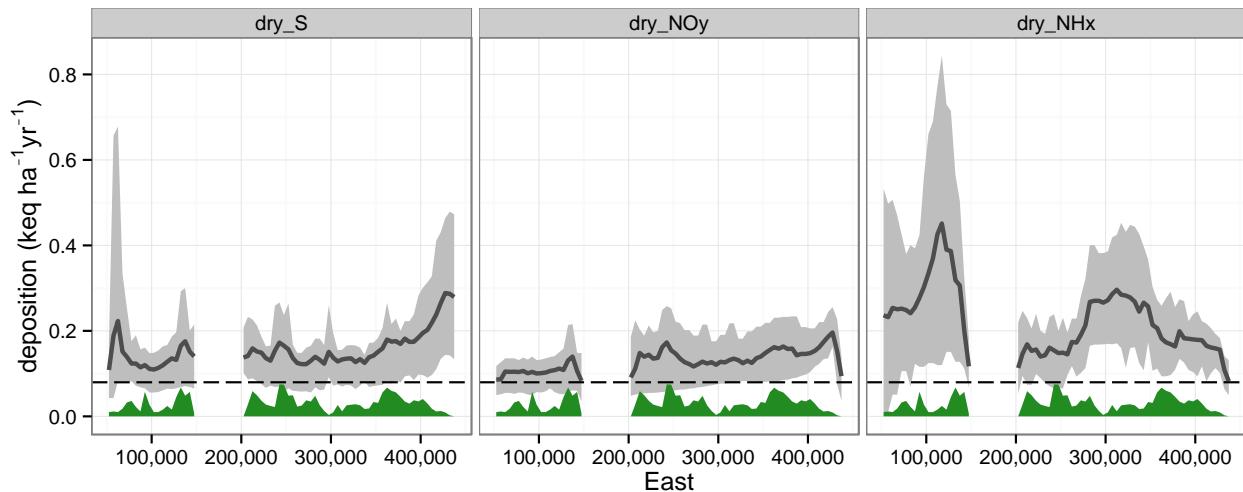
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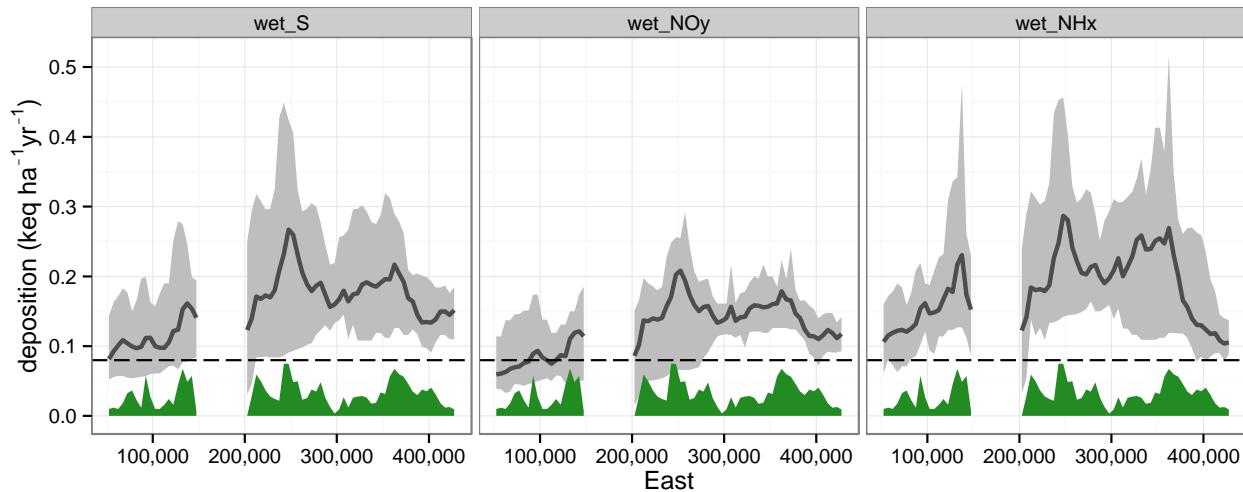
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467 magnitude of wet deposition at a local scale is found to vary considerably between models, by a  
 468 factor of up to 4 between the lowest and highest estimate.



469

470 **Figure 4(a):** Transect of dry deposition along a west-east trajectory (m) across the UK. The black line illustrates  
 471 deposition averaged for all models and the grey shaded area shows the range of minimum and maximum modeled  
 472 deposition. Terrain height is illustrated in green and the dashed line shows the 400m height.  
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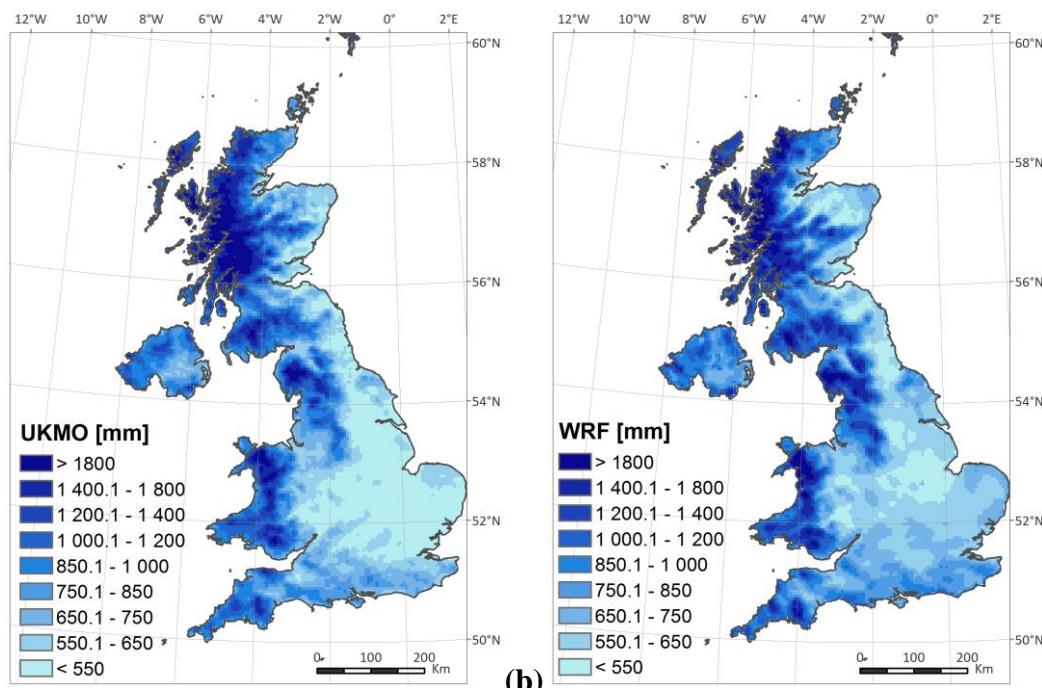
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476 **Figure 4(b):** Transect of wet deposition along a west-east trajectory across the UK. The black line illustrates deposition  
 477 across the UK. The black line represents the average deposition, and the grey shaded area shows the range of minimum and maximum modeled deposition.  
 478 Terrain height is illustrated in green and the dashed line shows the 400m height.  
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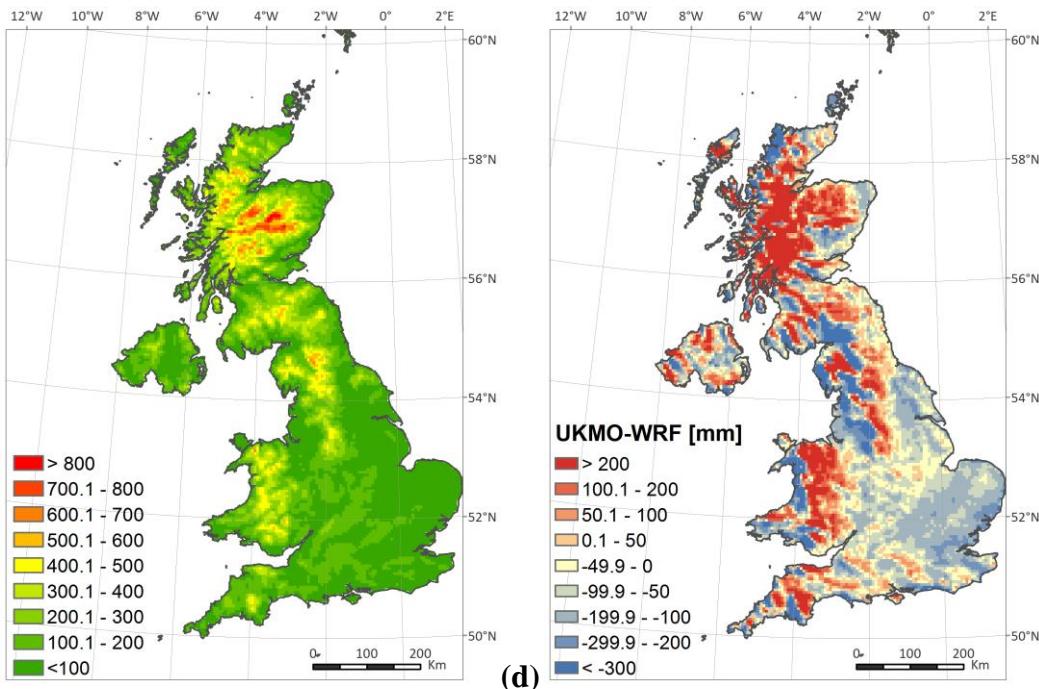
482 Variation in estimates of wet deposition can occur due to different precipitation values used  
 483 by the models. The simpler models (HARM and FRAME) use spatially distributed data based on  
 484 measurement and interpolation of annual precipitation measurements from the UK Meteorological  
 485 Office (UKMO) national precipitation monitoring network which is mapped at a 5 km resolution  
 486 (Simpson and Jones, 2012). Wet deposition in the complex models is driven by calculations of  
 487 dynamic precipitation from a meteorological model. For CMAQ and EMEP4UK the meteorological

488 driver is the WRF (Weather Research and Forecasting) model (<http://www.wrf-model.org>);  
 489 Skamrock and Klemp, 2008). The UKMO and WRF precipitation maps (Figures 5(a) and (b) )  
 490 show similar spatial distributions, with the lowest values of precipitation of approximately 600 mm  
 491 year<sup>-1</sup> along the east coast of England and the highest values, above 1800 mm year<sup>-1</sup>, in the hills of  
 492 Scotland, Wales and Northern England. High precipitation regions are closely correlated to terrain  
 493 height (Figure 5(c)). Analysis of the difference in precipitation between UKMO and WRF (Figure  
 494 5(d)) shows that the UKMO data generally have higher values in the upland areas, with differences  
 495 relative to WRF of over 200 mm year<sup>-1</sup>. Meteorological models may underestimate upland  
 496 precipitation due to the complexities of air flow and formation of cloud and rainfall in hill areas  
 497 which are not fully resolved at a 5 km grid resolution (Richard *et al.*, 2007). Precipitation in hill  
 498 areas is also uncertain using the UKMO measurement-interpolation approach. Rain gauges exposed  
 499 to higher winds and lower temperatures may capture precipitation inefficiently and fail to record  
 500 snowfall during sub-zero temperatures (Sevruk, 2009). Interpolation of rainfall in complex terrain  
 501 may introduce errors by failing to capture the influence of local orography on annual precipitation.  
 502 Hill areas are the regions of highest wet deposition and the location of sensitive ecosystems where  
 503 deposition of nitrogen may exceed critical loads but they are also the areas where precipitation is  
 504 less accurately estimated by both meteorological models and measurement-interpolation methods.

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507 **(c)**  Km **(d)** 

508 **Figure 5:** Annual precipitation for the year 2003 for (a) UKMO; (b) WRF;  
 509 (c) Terrain elevation for the UK; (d) Difference between UKMO and WRF precipitation

## 5. Discussion

The models were found on average to have negative mean biases for all precipitation concentration and gas and particulate phase measurements, except for SO<sub>2</sub>. It is important to review this in the light of systematic errors in measurement. Cape *et al.* (2009) used a ‘flushing sampler’ which, by detecting the onset of precipitation, was able to separately collect material dry deposited and that contained in precipitation. Comparison of this design with a standard bulk sampler over 3 months at a site in eastern Scotland showed that dry deposition to the funnel surface contributed approximately 20% of SO<sub>4</sub><sup>2-</sup>, 20–30% of NO<sub>3</sub><sup>-</sup> and 20–40% of NH<sub>4</sub><sup>+</sup> ions. Uncertainties in measurement of gas and aerosol concentrations may also occur due to incomplete reaction of gases with the substrate on a denuder tube, loss of aerosol mass by impaction on tubing, incomplete capture of fine particulate matter by filter papers and chemical analysis by ion chromatography. Araya *et al.* (2012) estimated an uncertainty of 20% in the measurement of anion and cation components in aerosol particles. It is therefore clearly inappropriate to set limits on the normalised mean bias of less than +/- 20% for model evaluation without due consideration of systematic errors in measurement technique. This emphasises the important issue that evaluation criteria to assess the performance of atmospheric chemical transport models should never be used alone in the absence of expert knowledge.

530        Wet only collectors are now widely used to collect samples of precipitation chemistry and can  
531        be combined with measurements of precipitation from tipping bucket rain gauges to give site-based  
532        estimates of wet deposition (i.e. van der Swaluw *et al.* 2011). Furthermore wet only collection is the  
533        standard procedure recommended by the World Meteorological Organisation. However there are  
534        technical issues (more complex maintenance; the need for electrical power; under-collection of  
535        precipitation) associated with their operation. Historically the reason for installation of a monitoring  
536        network for precipitation chemistry in the UK has not been for model validation but rather to detect  
537        long term trends in pollutant concentrations and provide adequate spatial coverage for mapping  
538        purposes. The emphasis has therefore remained on use of a simple low cost technique with a  
539        relatively dense network and continuity of measurement technique. However, the increased use of  
540        ACTMs in recent years to support national policy decisions inevitably means that the design of  
541        monitoring networks in the future should take account of their role in evaluating modelled  
542        concentrations.

543        The deposition model inter-comparison study was undertaken for the single year 2003 based on  
544        availability of input meteorological data for the Eulerian models. It is beyond the scope of the  
545        present study to include a detailed analysis of multiple years of data. However the question as to  
546        whether the results of the model comparison would have changed with the choice of a different year  
547        needs to be considered. Changes in annual circulation and precipitation can have a strong influence  
548        on concentrations of nitrogen and sulphur compounds in air and their deposition. Kryza *et al.*  
549        (2012) showed that inter-annual variability of precipitation and general circulation could cause  
550        major variations in sulphur and nitrogen deposition, equivalent to changes associated with long  
551        term emissions changes. 2003 was characterised by the lowest precipitation over the UK during the  
552        last 20 years. The 2003 UK annual average according to the UK Met Office precipitation maps was  
553        880 mm compared to 1130 mm averaged over the years 1986 – 2011  
554        (<http://www.metoffice.gov.uk/climate/uk/summaries>). Another feature of 2003 was the high value  
555        for the annual average aerosol concentrations in air. These were the highest for  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$  and  
556         $\text{NH}_4^+$  in air since measurements with the Delta samplers began in 1999 and approximately 50%  
557        higher than the long term average. Whilst the high aerosol concentrations were caused partly by  
558        low precipitation, a more important reason was the high incidence of south-easterly flow (in general  
559        an infrequent wind direction in the UK) leading to elevated concentrations of aerosol during the  
560        months of February, March and April caused by import of particulate matter from the European  
561        continent (Vieno *et al.*, 2014). 2003 may therefore be considered a somewhat uncharacteristic year  
562        for general circulation of air masses to the UK. A model may demonstrate good agreement with

563 measurements of, for example, total aerosol concentrations but not necessarily accurately capture  
564 specific atmospheric processes (i.e. the different relative contributions from national emission  
565 sources or long range transport). It cannot therefore necessarily be assumed that the correlation with  
566 measurements presented here for the year 2003 would be reproduced for other years with different  
567 meteorology and emissions. Simulating a year in which long range transport from the European  
568 continent made a greater than average contribution to sulphur and nitrogen concentrations poses  
569 additional challenges for ACTMs. Despite this fact both simpler and more complex models  
570 achieved a good degree of success in simulating the measured concentrations. A multiple year  
571 study is recommended for future work to assess model sensitivity to inter-annual changes in  
572 meteorology.

573 An alternative approach to emissions-based atmospheric chemical transport modelling is to  
574 make use of spatially distributed measurements combined with interpolation techniques to generate  
575 deposition data. This technique is frequently used to map wet deposition by combining  
576 measurements of concentrations of precipitation with annual precipitation measurements (i.e. Smith  
577 and Fowler, 2001). Spatially distributed dry deposition estimates can also be made by interpolating  
578 gas and aerosol concentrations and combining these with vegetation specific deposition velocities,  
579 as described by Smith *et al.* (2001) using a big leaf model. In the UK the combination of these dry  
580 and wet deposition estimates forms the Concentration Based Estimated (CBED) deposition data and  
581 has been used, averaged over three years, to estimate the exceedance of critical loads for nitrogen  
582 and acid deposition and changes in recent decades (RoTAP, 2012). For the year 2003 the CBED  
583 UK total annual deposition estimates for the UK showed significantly higher values for wet  
584 deposition (by 31%, 65% and 47% for SO<sub>x</sub>, NO<sub>y</sub> and NH<sub>x</sub> respectively) than the mean value of the  
585 ACTMs presented in this study. For dry deposition of NH<sub>x</sub> and SO<sub>x</sub>, CBED obtained similar values  
586 to the ACTMs. Dry deposition of NO<sub>y</sub> in CBED is currently under revision due to re-calibration of  
587 HNO<sub>3</sub> concentrations. The reasons for these differences require detailed investigation which  
588 should be undertaken in further work.

589 The model evaluation in this study has been based on annually averaged concentrations in air  
590 and precipitation because the simpler models are designed to calculate annual deposition and this is  
591 the standard data required for ecosystem impact assessment. More complex models are able to  
592 output data at high temporal resolution and can be subjected to a more detailed evaluation involving  
593 hourly or daily measured data. Future work should employ updated emissions estimates to calculate  
594 multiple year estimates of annual deposition of sulphur and nitrogen to the UK for use in

595 environmental impact assessments. It is recommended that this would include an assessment of the  
596 sensitivity of critical load exceedance (Hall *et al.*, 2006) to both choice of technique for deposition  
597 estimation (ACTM or measurements-interpolation system) and choice of individual model. More  
598 complex models are recommended as effective tools to assess future changes in nitrogen and  
599 sulphur deposition based on projected emissions reductions. The faster run times of simpler models  
600 however means that their application in studies requiring high resolution spatial simulation or large  
601 numbers of model runs (i.e. uncertainty estimates and source-receptor calculations) will continue to  
602 be useful.

603

## 604     **6. Conclusion**

605

606     An evaluation has been made of a range of simpler and more complex atmospheric chemical  
607 transport models, applied to make spatial estimates of acid deposition and nitrogen deposition to the  
608 UK. Deposition data from such models can be used to calculate the exceedance of critical loads  
609 which provides valuable information to policy makers on the need to reduce emissions of SO<sub>2</sub>, NO<sub>x</sub>  
610 and NH<sub>3</sub> to protect natural ecosystems. The models were evaluated by comparison with annually  
611 averaged measurements of gas (SO<sub>2</sub>, NO<sub>x</sub> and NH<sub>3</sub>), aerosol and precipitation concentrations (SO<sub>4</sub><sup>2-</sup>  
612 , NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>) from the national monitoring networks for the year 2003.

613     A model evaluation protocol was used to set the criteria for ‘fitness for purposes’. The first  
614 condition, that at least 50% of modelled concentrations should be within a factor of two of the  
615 measured value, was generally satisfied by the models. The second criteria, that the magnitude of  
616 the normalised mean bias should be less than 20%, was not always satisfied. Uncertainties resulting  
617 from measurement techniques were not accounted for in this analysis, however these can be  
618 significant. In particular ion concentrations in precipitation can be overestimated by 20-40% using  
619 bulk collection of precipitation samplers (Cape *et al.*, 2009). It is therefore recommended that  
620 uncertainties and biases in measurement technique are taken into account when using model  
621 evaluation criteria to judge whether a model is fit for purpose. For example, ‘adjusted NMB’  
622 criteria could be used which had variable maximum and minimum limits of acceptability for the  
623 normalised mean bias that were dependent on errors and uncertainty in measurement techniques.

624     Simple models have practical advantages due to their fast run times and ability to perform  
625 multiple simulations and performed satisfactorily when compared with measurements. Complex  
626 models are able to more accurately represent chemical transformation and long range transport of  
627 pollutants leading to better representation of particulate concentrations of SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>.

628 They also benefit from the ability to simultaneously represent other pollutants (i.e. surface ozone).  
 629 No attempt was made to rank the models overall. However it was clear from the evaluation that  
 630 different models performed best for different pollutants (sulphur, oxidised nitrogen, reduced  
 631 nitrogen) and states (gas, particulate, aqueous) so that in practical terms ranking would not be a  
 632 simple task.

633 Comparison of the modelled deposition budgets to the UK showed that total deposition  
 634 varied by +/- 22-36 % depending on model deposition parameter, with similar variability amongst  
 635 both wet and dry deposition estimates. At a local (5 km grid square) scale however, variability in  
 636 estimates of deposition amongst models could be very much higher, varying by up to a factor of  
 637 four between different models. These results give an indication of the uncertainty associated with  
 638 estimating sulphur and nitrogen deposition due to choice of model. Variation, and therefore  
 639 uncertainty, was notably high for wet deposition in high precipitation upland areas, regions where  
 640 ecosystems which are sensitive to nitrogen deposition are present.

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#### 644 Acknowledgement

645

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 647 support was provided by the Joint Environmental Program, the Natural Environment Research  
 648 Council and the Environment Agency.

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