- **1** Informal E -Waste Recycling in Nine Cities of Pakistan Reveals Significant Impacts on Local
- 2 Air and Soil Quality and Associated Health Risks
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31 Abstract

The global increase in electronic waste (e-waste) has led to a rise in informal recycling, emitting 32 hazardous heavy metals (HMs) that threaten human health and ecosystems. This study presents 33 the first comprehensive assessment of HM levels in dry deposition and soils at proximity of forty 34 35 (40) informal e-waste recycling sites across Pakistan, between September 2020 to December 2021. 36 Findings reveal that Zn (1410), Pb (410) and Mn (231) exhibited the higher mean deposition fluxes (µg/m².day), derived from air samples, particularly in Karachi. Similarly, soils showed higher 37 38 mean concentrations (µg/g dw) of Mn (477), Cu (514) and Pb (172) in Faisalabad, Lahore, and 39 Karachi, respectively. HMs concentrations were found higher in winter or autumn and lower in 40 summer. In addition, HM levels were significantly (p=0.05) higher at recycling sites compared to 41 background sites year-round, highlighting the e-waste recycling operations as the major source of their emissions. The Igeo index indicated moderate to extremely contaminated levels of Cu, Pb, Cd, 42 and Ni in Karachi, Lahore and Gujranwala. Ingestion was found as a leading human exposure 43 44 route, followed by dermal and inhalation exposure, with Pb posing the greatest health risk. The Cumulative Incremental Lifetime Cancer Risk (ILCR) model suggested moderate to low cancer 45 risks for workers. Strategic interventions recommend mitigating health and environmental risks, 46 prioritizing human health and ecosystem integrity in Pakistan's e-waste management. 47

48 Keywords: E-waste; Informal recycling; HMs; Particulate matter; Soils; Health risks

49 **1. Introduction**

50 Due to rapid technological advancements electronic and electrical equipment (EEE) production 51 and use has experienced exponential growth in the last two decades which in turn has led to a rise in electronic and electrical waste (e-waste) (Kumar et al., 2017; Murthy et al., 2019). E-waste 52 53 denotes to the end-of the lifespan of electrical and electronic products such as, televisions mobile phones, computers with plastic as major constituent and contain heavy metals (HMs) like 54 Manganese (Mn), Cadmium (Cd), Copper (Cu), Lead (Pb), Chromium (Cr), Zinc (Zn), Nickel (Ni) 55 as well as other toxic compounds e.g., polychlorinated biphenyls (PCBs) and polybrominated 56 diphenyl ethers (PBDEs) (Brindhadevi et al., 2023; Qin et al., 2022). 57

As per recent data (2019) the global production of electronic waste reached 53.6 million metric 58 tons (mt), equating to 7.3 kilograms per capita annually, up from 5.8 kilograms per capita in 2014 59 (Houessionon et al., 2021; Forti et al., 2020). E-waste has emerged as a significant waste stream 60 worldwide, with its generation rate increasing by 1 to 5% annually (Kumar and Fulekar, 2019). 61 Typically, developed countries either dispose of their e-waste in landfills or export it to developing 62 nations under the title of second-hand item or recycling (Dutta et al., 2022; Arya et al., 2021). E-63 64 waste recycling facilities have been reported in many countries of the world, like China, India, Pakistan, the Philippines, Turkey, and Vietnam (Hashmi et al., 2022). It is estimated that 65 developing nations receive approximately 80% of the world's e-waste (Sthiannopkao & Wong, 66 2013), and up to 2% of the Asian population relies on e-waste recycling for their livelihood (Imran 67 et al., 2017). In developing countries, cheap labor force, weak legislation, and a substantial market 68 for recovered materials make them a target and vulnerable destination for the informal recycling 69 and dumping of e-waste (Singh et al., 2018). Informal recycling practices include acid baths for 70

the recovery of gold and other valuable metals, stripping/shredding and open-air burning, grilling, 71 chipping, and melting plastics, disposing of unusable material in open fields and water bodies. 72 Through these activities, several hazardous elements such as Pb, Cd, Ni, Cu, and other metals can 73 be emitted into the surroundings. Thus, exposure of these toxic elements to humans and their 74 effects are of great concern (Song and Li, 2014). For example, exposure to Pb can affect the liver, 75 76 kidney, and nervous system and impair cognitive development (Obeng-Gyasi, 2018; Bellinger, 2011). Cr can cause respiratory irritation, kidney, and liver damage, weakened immune systems, 77 and cancer of the nose, sinus, or lung (Tchounwou et al., 2012). Ni may contribute to dermatitis 78 79 and bronchial asthma (Kuntawee et al., 2020).

Pakistan has been documented as one of the major e-waste importers; yet, there is no quantification 80 of the e-waste inventory flows (Iqbal et al., 2017). Pakistan also generates a considerable amount 81 of e-waste with nearly 433kt of e-waste was produced in the year 2019 as compared to 301kt in 82 2016, showing a 43.8% rise and attaining it the 20th biggest e-waste producer (Forti et al., 2020). 83 In Pakistan, e-waste primarily originates from two main sources: legal/illegal import of electrical 84 and electronic equipment and domestic generation (Iqbal et al., 2015). Once discarded, sellers and 85 scrappers collect and disassemble the waste into various parts. Dismantlers and extractors make 86 87 their way by treating such waste using different informal methods. Informal techniques such as open burning, acid baths, physical dismantling and treatment using blow torches are employed by 88 89 these dismantlers and recyclers to recover valuable metals. Notably, Pakistan lacks formal e-waste 90 recycling facilities and entire e-waste is processed through informal channels (Hameed et al., 2020). Driven by profits, this extensive informal and illegal recycling of e-waste operates at the 91 92 stake of human and environmental health.

Previously, several studies highlighted the close correlation between human exposure and e-waste 93 recycling (Arya et al., 2021; Mowla et al., 2021; Perkins et al., 2014; Pradhan and Kumar, 2014; 94 Shakil et al., 2023). Numerous studies have reported HMs in various environmental matrices in 95 Pakistan (Saleem et al., 2018; Khan et al., 2015; Mahmood and Malik, 2014), however only one 96 study reported soil contamination from e-waste dumping and recycling sites in Lahore (Shakil et 97 98 al., 2023). As per the reported literature, this is the first comprehensive research in Pakistan that provides the temporal and/or seasonal trends of HMs in ambient air and soils, and their exposure 99 risks covering an extensive sampling campaign across Pakistan. 100

101 2. Materials and methods

102 **2.1.** *Study Area Description and Sampling Scheme*

Informal e-waste recycling sites were determined through the following methods: (1) conducting 103 field surveys to locate sites involved in e-waste repair, dismantling, and refurbishment; (2) 104 identifying national entry points; and (3) referencing previous studies (Shaikh, 2021; Sajid et al., 105 2019; Imran et al., 2017; Igbal et al., 2017, 2015; Umair et al., 2016). A total of 40 e-waste 106 recycling facilities through nine major urbanized cities in Pakistan were selected for sampling. 107 These cities include Rawalpindi, Faisalabad, Lahore, Gujranwala, and Multan (Punjab); Karachi 108 109 and Hyderabad (Sindh); Quetta in Baluchistan; and Peshawar in Khyber Pakhtunkhwa (KPK) province. Additionally, the COMSATS University Campus in Islamabad was selected as the 110 111 background site (refer to Figure 1). Further details, such as site names, codes, geographical coordinates, and descriptions for each sampling location, are summarized in Table S1. 112

Overall, 160 Passive air deposition samplers (PASs-DD) and 160 soil samples were collected for assessing the atmospheric load and soil residues of heavy metals in four seasons (autumn, winter, spring, and summer) between September 2020 and December 2021. Details about the deployment of PASs at each sampling location are given in Table S2, while meteorological conditions are presented in Table S3. To collect ambient particulate deposition in air, PASs-DD were deployed at varied distances from e-waste recycling sites but all within a radius of ~ 200 m. Deployment time of PASs-DD varied between 62 to 135 days with an average of 100-days. Samplers were deployed on the rooftops of buildings at ~10 m height, to minimize the influence of re-suspended soil dust which can occur near ground level.

122 **2.2.** *Sampling*

123 **2.2.1.** Soil samples collection

A hand-held corer was used to collect soil samples between 0-10 cm depth. At least 10 cores were collected randomly at individual sampling locations and a composite sample was obtained after mixing all-together. Soil samples collected were sieved through a 2 mm mesh metal sieve. Moisture and organic content in soil samples were determined according to ASTM D-2974-87 (ASTM, 2000) method and soil pH measurements were made according to EPA method 9045-D (US EPA, 2004). The detail of physicochemical analysis is presented in supporting Information (Table S4).

131 **2.2.2.** *Atmospheric Particle-bound Heavy Metals*

Sampling of air deposition, which for heavy metals is mainly associated with particulate matter, was performed by deploying passive dry deposition (PASs-DD) collectors using a polyurethane foam disk (PUF-disk). PASs-DD have successfully been used to determine the flux of polycyclic aromatic hydrocarbons (PAHs) (Eng et al., 2014) and HMs (Gaga et al., 2019) in urban environments and collect both particle-phase dry deposition and gas-phase. The information about design and sampling rate of -PAS can be found elsewhere (Gaga et al., 2019; Eng et al., 2014). To the best of our knowledge, this is the first research to employ PASs-DD to map the atmospheric fluxes of HMs in and around Pakistani e-waste recycling facilities. PUF disks were precleaned prior to deployment in accordance with the standard clean-up procedure applied for persistent organic pollutants (POPs) analysis (Birgul and Karakus, 2024). After harvesting at the sampling locations, PUF disks were tightly closed in aluminum bags, then placed within zip-lock bags and a locked container for transport to COMSATS University Islamabad. At the university, they were stored in a dry, clean room until they could be shipped to Bursa Technical University in Turkey for sample preparation. Prepared samples were analyzed for HMs in Dokuz Eylul University.

146 **2.2.3** Sample Preparation

Weight of each PUF disk was recorded before and after deployment in the field, hence approx. 147 dust amount collected was determined (0.04 g-4.39 g). Subsamples taken out of each PUF disks 148 that were deployed for approx. 3 months at the sampling sites were used to analyze heavy metals. 149 A stainless-steel corer was used to cut 1 cm diameter cores from 5 randomly selected points on 150 each disk to obtain sub-samples. The PUF disk was weighed again after 5 cores ($\Sigma 0.11$ g) were 151 152 taken out to determine the weight of the cores as well as weight of dust collected in these 5 cores (0.02 g-0.22 g). Wet digestion method was applied to prepare the samples for instrumental analysis. 153 The subsample consisting of 5 cores were placed in a 40 mL glass vial, 2 mL of H₂SO₄ and 6 mL 154 155 of nitric acid HNO₃ and were added into the vial. Vials containing PUF disk subsamples and acid mixture were placed on an aluminum heating block and digestion was carried out by heating 156 samples at temperatures ranging from 160 °C to 180 °C until the formed brownish fume 157 disappeared and the solution became clear. After the digestion was completed, the samples were 158 kept in the fume hood until they are cooled to room temperature and diluted to 50 mL using 159 ultrapure water. To remove any impurities an aliquot of 15 mL of the digested sample was filtered 160 through a 0.45 µm pore size Teflon syringe filter. Filtered samples were placed in 15 mL volume 161

plastic falcon tubes, the caps of the tubes closed tightly and refrigerated at -18 °C until instrumental
analysis.

164 **2.2.4.** *Instrumental Analysis*

Analysis of heavy metals was conducted using an Inductively Coupled Plasma-Mass Spectrometer 165 (ICP-MS) (Agilent 7700x). The operational parameters of the device were as follows: RF power 166 167 set to 1550 W; carrier gas flow rate at 0.90 L/min; plasma gas flow rate at 15 L/min; plasma sampling depth of 8 mm; nebulizer flow rate set at 1.01 L/min; extractor lens potential at -160 V; 168 169 conical spray chamber temperature maintained at 2°C; nebulizer pump operated at 0.10 cycles/second; nebulizer type used was micro-mist; and the ion lenses model employed was x-170 lens. Readings were taken as three replicates and average value of three readings were used as the 171 concentration value of a sample. 172

173 **2.3.** *Quality Assurance / Quality Control (QA / QC)*

Accurate quality assurance and quality control practices were pursued from sample collection 174 through to analysis. All laboratory equipment was either high quality polypropylene or Teflon. All 175 chemicals were analytical grade. Field blanks were brought to the laboratory in closed boxes/bags 176 after being exposed to ambient air for 1-2 minutes. A mixture of acids which were used for sample 177 digestion was used in preparation of laboratory blank samples. A total of 12 blank samples for 178 PUFs and a total of 14 blank samples for soils were prepared and they were handled in the equal 179 180 manner as the samples. Reproducibility of the obtained results were checked by analyzing CRM 540. Relative standard deviation of results between certified value and analysis ranged between 181 1.38% (Mn) and 14.4% (Pb) with an average RSD of 6.31±4.41%. Further details on RSD values 182 183 detected for target contaminants are given in Supporting Information Table S5. For digestion, method spike samples (n=10) were prepared by adding known quantity of target elements (100 184

ppb each) into acid mixture and process was carried out similar manner as done for samples. The 185 average recovery ratio was 96.4±4.89% ranging between 89.7% (Co) and 103% (As and Pb). The 186 Instrument Detection Limit (IDL) was based on half the concentration value of the lowest 187 calibration level that the instrument was able to determine in the set of calibration solutions. 188 Samples were not blank corrected. Therefore, method detection limit (MDL) was calculated based 189 190 on average blank concentration + 3 x Standard deviation (SD) of the concentration detected in the blank samples. IDL and MDL values of each targeted element are given in Supporting Information 191 Table S6. 192

193 2.4. *Calculation of Particulate Phase Pollutant Fluxes*

194 Fluxes were presented as mass per unit area per unit time (1) (i.e., $\mu g/m^2$.day) in Eq. 1 as follows:

195 Deposition Flux (F) =
$$\frac{m}{t} \times A$$
 (1)

Where m is the quantity of HMs determined in the PAS-DD sample (μ g), t is the total deployment duration of the sampler in the field (days), A is the surface area where deposition of particles occurred (0.00785 m², as edges and bottom of the DD-PAS sampler sampling medium holder unit is relatively closed, therefore it was assumed that deposition occurred mainly on to the top side of PUF disk).

201 **2.5.** *Geo-accumulation Index* (Igeo)

The contamination level in soils from this study were determined by calculating the values of the geo-accumulation index (I_{geo}) (also known as Muller index) (Han et al., 2018; Muller, 1981). This indicator is used to calculate the scale of contamination by assessing the relation between calculated concentration level and background level of the contaminant (Muller, 1969). Eq. (2) is used to calculate I_{geo}:

$$207 \quad I_{\text{geo}} = \text{Log2} \times \frac{\text{C}}{1.5\text{BG}} \tag{2}$$

Here C represents the heavy metal concentrations in the soil samples analyzed while BG is the geo-chemical background concentration of the element in the earth's crust and 1.5 is background matrix correction factor as consequence of lithogenic effects. This coefficient depicts any anthropogenic effect in the computation as well as the influence of geological and depositional features.

213 **2.6**. *Enrichment Factor (EF)*

Assessing the contamination level of heavy metals from human activities involves comparing the concentrations of heavy metals in soil and particulate samples with those of reference elements found in the Earth's crust. For this purpose, enrichment factor (EF) is calculated as shown in the Equation (3) below (Al-Khashman, 2013; Abdulaziz et al., 2022; Tepe et al., 2022) for soil and particulate matter, separately.

219
$$EF = \frac{\left(\frac{C_i}{C_{ref}}\right)_{sample}}{\left(\frac{C_i}{C_{ref}}\right)_{earth \ crust}}$$
(3)

220 Where Ci is the concentrations of target heavy metals $(\mu g/g)$ in particulate matter or soil and Earth 221 crust. Cref represents reference elements concentration in particulate matter/soil and Earth crust. 222 As Mn is one of the reference elements used in previous studies (Rahman et al., 2021; Pasha et al., 223 2015; Zajusk-Zubek et al., 2015; Fabretti et al., 2009; Sakata and Asakura, 2011; Abdulaziz et al., 224 2022), it is used as reference element in the current study. Tayler (1964) provides Heavy metals 225 concentrations in Earth's crust. An EF level of ≤ 10 indicates the cause of the metal from the natural source of Earth's crust while EF value of >10 suggests anthropogenically enriched (Duan et al., 226 227 2021; Kodat et al., 2023)

228 **2.7.** *Human health risk assessment*

Exposure of HMs to the human body can occur through ingestion via mouth, inhalation via mouth and nose, and dermal exposures via skin when in proximity of informal e-waste recycling. The present study calculates the non-carcinogenic health and lifetime cancer risk based on inhalation, ingestion, and dermal exposure routes of HMs. In addition, overall data used for the calculation of average daily intake is given in Table S7.

234 2.7.1 Estimation of Daily Intake Through Contaminated Soil

The potential risk from heavy metals in contaminated soils collected from informal e-waste recycling sites are calculated based on recommendations proposed by United States Environmental Protection Agency (USEPA 1989; 1997; 2000 and 2001). The average daily intake (ADI) (mg/kgday) of each heavy metal through soil ingestion, inhalation and dermal contact pathways was calculated using the following equations (4-5) (Ajani et al., 2022):

240
$$ADI_{Soil-Ing} = \frac{C_{soil} x IngR_{soil} x EF x ED}{BW x AT x 10^6}$$
 (4)

241
$$ADI_{Soil-Inh} = \frac{C_{soil} x InhR x EF x ED}{BW x AT x PEF}$$
 (5)

242
$$ADI_{Soil-Dermal} = \frac{C_{Soil} \, x \, SA \, x \, FE \, x \, AF \, x \, ABS \, x \, EF \, x \, ED}{BW \, x \, AT \, x \, 10^6}$$
 (6)

Where ADI_{Soil-Ing}, ADI_{Soil-Inh} and ADI_{Soil-Dermal} are the average daily intake doses through soil ingestion, inhalation, and dermal absorption, respectively (mg/kg/day) and C_{soil} is the concentration of heavy metal in soil.

246 **2.7.2** *Estimation of daily Intake through air particulate matter*

Human exposure is also measured in terms of average daily intake (ADI) via ingestion (Eq. 7) and dermal absorption (Eq. 8) (mg/kg/day) of air particulate matter and exposure concentration via inhalation (EC) of air particulate matter (Eq.9) (Abdulaziz et al., 2022).

$$250 \quad ADI_{PM-Ing} = \frac{C_{PM} x \ln g R_{PM} x EF x ED}{BW x AT x 10^6}$$
(7)

$$251 \qquad EC_{PM-Inh} = \frac{C_{PM} \, x \, ET \, x \, EF \, x \, ED}{AT_n} \tag{8}$$

252
$$ADI_{PM-Dermal} = \frac{C_{PM} x SA x AF x ABF x EF x ED}{BW x AT x 10^6}$$
(9)

Where ADI_{PM-Ing}, and ADI_{PM-Dermal} are the average daily intake doses through particulate matter ingestion and dermal absorption, respectively (mg/kg/day) and EC_{PM-Inh} is the exposure concentration via inhalation (μ g/m³) of air particulate matter. C_{PM} is the concentration of heavy metal (mg/kg for ADI_{PM-Ing}, and ADI_{PM-Dermal}; mass per unit volume (μ g/m³) for EC_{PM-Inh}).

257 2.7.3 Non-Carcinogenic Health Risk

Non-carcinogenic adverse health effects assessment is carried out for both carcinogenic and non-258 carcinogenic heavy metals. The IARC has classified As, Cr, Cd and Ni as Group 1 carcinogens, 259 whereas Pb and Co were classified as Group 2A carcinogens (IARC, 2024). USEPA (1989; 1997; 260 2000 and 2001) proposed that target hazard quotients (HQ) and hazard index (HI) characterize the 261 potential health risk. HQ is a ratio of determined average daily intake (ADI, (mg/kg/day)) to 262 reference dose (RfD, (mg/kg/day)) of an individual element. HQ values ≤ 1 indicate no significant 263 264 or acceptable risk, while HQ values >1 indicate the potential for adverse health effects (USEPA, 2001). For a given heavy metal, HQ values for exposure through soil ingestion, inhalation, and 265 dermal contact in addition to particulate matter ingestion and dermal contact is calculated using 266 267 the equations given below (Ajani et al., 2022; Abdulaziz et al., 2022).

$$268 \qquad HQ_{Soil-Ing} = \frac{ADI_{Soil-Ing}}{RFD} \tag{10}$$

$$269 \qquad HQ_{Soil-Inh} = \frac{ADI_{Soil-Inh}}{RFD} \tag{11}$$

270
$$HQ_{Soil-Dermal} = \frac{ADI_{Soil-Dermal}}{RFD}$$
 (12)

$$271 \qquad HQ_{PM-Ing} = \frac{ADI_{PM-Ing}}{RFD}$$
(13)

272
$$HQ_{PM-Dermal} = \frac{ADI_{PM-Dermal}}{RFD}$$
 (14)

273 Whereas RfD is the reference dose (mg/kg/day) (Table S8).

HQ value for exposure through particulate matter inhalation can be calculated based on the
equation given below (Abdulaziz et al., 2022)

$$276 \qquad HQ_{PM-Inh} = \frac{ADI_{PM-Inh}}{RfCx1000} \tag{15}$$

277 Whereas RfC is the reference concentration of the heavy metal (mg/m^3) (Table S8)

278 Health risks associated with exposure to multiple metals is estimated by using Hazard index (HI)

- 279 (the summation of hazard quotients (HQ_k) of individual metal "k") which can be calculated using
- the following equation (USEPA 2001; Khan et al., 2020)

$$HI = \sum HQ_k \tag{16}$$

For non-carcinogenic or carcinogenic heavy metals, a value of HI > 1 represents that there is a chance of occurrence of non-carcinogenic effects, while the exposed individual is unlikely to experience obvious adverse health effects when HI < 1.

285 **2.7.4** *Lifetime Cancer Risk*

The probability of developing cancer because of human exposure to carcinogenic heavy metals (As, Cr, Cd and Ni as Group 1 carcinogens and Pb and Co as Group 2A carcinogens (IARC, 2024)). Cancer risk over the lifetime (ILCR) can be estimated using equations below for ingestion, inhalation, and dermal contact respectively.

290
$$ILCR_{\sum ing} = (ADI_{Soil-Ing} + ADI_{PM-Ing}) \times SF$$
 (17)

291
$$ILCR_{\sum inh} = (ADI_{Soil-Inh} x SF) + (EC_{PM-Inh} x IUR)$$
 (18)

292
$$ILCR_{\sum dermal} = (ADI_{soil-dermal} + ADI_{PM-dermal})x SF$$
 (19)

Where ILCR_{Σ ing}, ILCR_{Σ inh}, ILCR_{Σ dermal} represents incremental lifetime cancer risks via 293 soil+particulate matter ingestion, inhalation, and dermal contact, respectively. Values of Slope 294 factor (SF, mg/kg.day) and inhalation unit risk (IUR, $\mu g/m^3$) for carcinogenic metals are given in 295 Table S8. Classification of ILCR is as follows: ILCR $\leq 1 \times 10^{-6}$ (very low); $10^{-6} \leq ILCR \leq 10^{-4}$ (low); 296 10^{-4} < ILCR < 10^{-3} (moderate); 10^{-3} < ILCR < 10^{-1} (high) and ILCR > 10^{-1} (very high) (Zhang et al., 297 2021). The Cumulative ILCR for a given carcinogenic metal can be calculated as the sum of ILCR 298 values occurred due to ingestion, inhalation, and dermal contact (Sun et al., 2021) and this value 299 should be maintained below 10⁻⁴ (Chalvatzaki et al., 2019). 300

301 Cumulative ILCR =
$$ILCR_{\Sigma ing} + ILCR_{\Sigma inh} + ILCR_{\Sigma dermal}$$
 (20)

302 **2.8.** *Statistical analysis*

IBM SPSS Statistics software (version 20.0) was used for statistical analysis. One-way repeated
measures multivariate analysis of variance (RMANOVA) was applied to determine significance
of differences in HMs concentration over four seasons in each city. Arc-GIS software (version
10.2.2) was used for site identifications at all sampling sites across nine cities.

307 3. Result and Discussion

308 3.1. *Concentrations levels of Heavy metals at e-waste sites*

Mean level of HMs in air and soil as well as background site in four seasons are detailed in Table 309 1. At background site in Islamabad, the average deposition flux of HMs in air during four 310 deployment seasons was noted as $41.8 + 33.8 \,\mu\text{g/m}^2$.day. Whereas the average deposition flux was 311 found to be $161 + 111 \,\mu\text{g/m}^2$.day ranging from 56.0 $\mu\text{g/m}^2$.day (Rawalpindi) to 331 $\mu\text{g/m}^2$.day 312 (Karachi). Substantially, higher deposition flux at study sites (> 3 times the background site, Table 313 1) presumably due to the presence of active sources of HMs at studied sites. Deposition flux of 314 HMs in air was found in following sequence Zn > Mn > Pb > Cu > Ni > Cr > Cd. Since, the present 315 study is first of its kind from Pakistan in which passive samplers were deployed to study HMs and 316 report concentration in the units of flux ($\mu g/m^2$.day), therefore the comparison of deposition flux 317 of HMs in air was not viable with regional or global studies using different methodologies. The 318 compositional trend of HMs in present study were accorded well with those of other e-waste 319 320 recycling sites in previous studies with higher concentrations of Pb, Zn and Mn than Cr, Ni, Cu and Cd (Table S9). The elevated levels of HMs at study sites compared to the background site 321 indicate the contribution of emissions from e-waste recycling sites to the local atmosphere. 322

For soil samples, the mean concentrations from recycling facilities were much greater (~1.5 to 13 times) than the background concentration suggesting the influence of extensive e-waste recycling operations (Table 1). The mean concentrations of HMs in soils were found in the following sequence Mn>Zn>Cu>Pb>Cr>Ni>Cd which is almost identical to those reported for previous studies at e-waste recycling, dumping and/or dismantling sites (Table S10). This shows that ewaste dismantling and recycling activities substantially contribute towards the contamination of soil. The mean concentration and their ranges for Mn, Cr and Ni were comparable to those noted

in Nigeria (Isimekhai et al., 2017), Ghana (Teye et al., 2023), and China (Han et al., 2019), whereas 330 for Cu, Zn, Cd and Pb, the concentrations were much lower than reported by those studies (Table 331 S10). Soil pollution is often assessed either by comparing total metal concentrations with standard 332 guideline values or by classifying using pollution indices (Wu et al., 2018). In this study, it is 333 apparent that the mean concentrations of Zn and Cd in soils from e-waste facilities exceeded the 334 335 safe regulatory limits of WHO i.e., 50 μ g/g (Osobamiro et al., 2019) and 0.003 μ g/g (Ahmad et al., 2021), respectively. Out of nine selected cities, mean concentrations of Ni, Cu and Pb in four 336 major industrial cities *i.e.*, Karachi, Lahore, Faisalabad and Gujranwala were higher than WHO 337 standard limits of 35 µg/g (El-Naggar et al., 2021), 100 µg/g and 50 µg/g (Teye and Tetteh, 2023), 338 respectively. The soil concentration of Cr in all cities was found under the standard limit of WHO 339 *i.e.*, 100 μ g/g (Teye and Tetteh, 2023) except for Gujranwala city where the concentration (112) 340 $\mu g/g$) just exceeded the standard limit. For Mn, except in Lahore and Faisalabad, its mean levels 341 were within the recommended limits of WHO (437 μ g/g) (Bawwab et al., 2022). 342

343 Heavy metal's deposition fluxes reported in the current study were generally align with the patterns observed in the earlier research conducted from various parts of the world, reinforcing the global 344 nature of the issue. For instance, similar to findings in India (Ha et al., 2009), Nigeria (Isimekhai 345 346 et al., 2017), China (Han et al., 2019 and Ghana (Teye et al., 2023), Pakistani cities exhibit elevated levels of Zn, Mn, Pb, Cu, Ni and Cr as these metals are often associated with electronic waste 347 348 (Purchase et al., 2020). Zn is used in die-castings, batteries fluorescent lights and X-ray screens in 349 EEE while Mn as an alloy with Pb, Al or Cu is a major constituent of batteries, sensors, and super capacitors. Similarly, Al, Pb Cr, and Cu are major components of printed circuit boards, smart card 350 351 chips, electrical wiring, and various other EEE. During recycling process (mostly informal) 352 including dismantling, repairing, burning and acids treatment to recover precious metals may

possibly initiate heavy metal contamination at the e-waste recycling facilities (Li et al., 2011). In addition to the overall concentration of the metals, the extent of contamination is determined by the fraction of their movable and bioavailable forms, which in general controlled by the organic matter, pH and other properties in soil (Tang et al., 2010).

357 3.2. *Spatial trends*

358 An intra-city relationship of HMs in the air samples near e-waste facilities is illustrated in Figure 2 and S1 while descriptive statistics is given in Table S11.. Karachi, Gujranwala, Lahore, and 359 360 Faisalabad were found to be the leading cities with higher level of HMs. Out of studied HMs, four metals were found higher in Karachi (µg/m².day) i.e., Cr (with annual mean 38.4), Mn (231), Zn 361 (1410) and Pb (410) whereas higher fluxes of Ni (157) and Cu (255) were noted for Gujranwala. 362 In Karachi, the maximum fluxes (µg/m².day) of Cr (131), Mn (1520), Ni (276), Cu (931), Zn 363 (8105), Cd (23.4), and Pb (2993) were observed at Sher shah (J18) or its adjacent Lyari area (J20). 364 Sher Shah market is the biggest junkyard of Pakistan for used electronic and locomotive parts 365 storage, dismantling, and recycling reported in earlier studies (Hameed et al., 2020; Rafeeq et al., 366 2021). Recently, Kazim et al. (2023) also reported the higher levels of gaseous elemental mercury 367 at this site which they had associated with e-waste dismantling and recycling processes in Pakistan. 368 369 Among the sampling cities, HMs concentrations in Rawalpindi, Multan, Quetta, and Peshawar were lower but still well above background HM levels. High variability of Zn, Cu, Pb, Mn and Ni 370 in major cities in comparison to background concentrations (Table 1) may also be due to different 371 sources in addition to contribution from e-waste recycling sites. These sources could also be 372 attributed to the industrial emissions and vehicular traffic (Zhou et al., 2014) as most of the studied 373 sites are densely populated and industrial hubs of some scale. Vehicles emit HMs into the 374 atmosphere mainly via exhaust (fossil fuel emissions) and non-exhaust emissions which include 375

wearing and tearing of different vehicular sections like tires, brake pads, and corrosion of metallic
parts. Therefore, USEPA highlights 21 hazardous elements that can mostly be appointed to road
traffic (Gupta, 2020); and the five dominant HMs (Zn, Cu, Pb, Mn and Ni) as observed in our
study are among them.

Descriptive statistics for the HMs concentrations found in samples collected from soil at 380 381 contaminated sites in nine cities are summarized in Table S12 and their variations are shown in Figure 3, while spatial distribution maps are given at Figure S2. Higher mean concentrations ($\mu g/g$ 382 dw) of Cr (112), Ni (79), Cu (457) and Cd (1.32) were found in Gujranwala. Among four selected 383 study sites in Gujranwala, the industrial zone site (J26 Site) had major contribution in elevating 384 the mean concentration of HMs. Maximum concentrations of Cr, Ni, Cu, and Cd at this site were 385 observed up to 716, 542, 2565 and 4.53 (in $\mu g/g dw$) respectively in different seasons which were 386 nearly $1 \sim 2$ orders of magnitude higher than the background site (Table 1). The city of 387 Gujranwala's industrial zone is a center for buying/selling of e-waste with informal recycling of 388 389 printed circuit boards (PCBs) by using include acid baths, open burning etc. Highest mean concentrations of Mn (477 µg/g), Cu (514 µg/g) and Pb (172 µg/g) in soil were detected in 390 Faisalabad, Lahore, and Karachi. The dominant site in Faisalabad, which had particularly 391 392 contributed to mean concentrations of HMs was Motor Market (J9) where the concentration of Mn (1415 µg/g dw), Cr (142 µg/g dw), Zn (1014 µg/g dw) and Pb (221 µg/g dw) were higher than 393 394 other three selected sites within the city. In Lahore and Karachi, the elevated inter-city 395 concentrations of mostly HMs were observed in soil samples collected from Misri Shah (J15) and 396 Sher Shah (J18) respectively. The description of all these sites is summarized in Table S1.

397 *3.3.* Seasonal trends

Previous studies have reported substantial influence of meteorological conditions (temperature, 398 wind speed, rainfall) on levels of air pollutants (Nasir et al., 2019) and physicochemical 399 characteristics of soils (Aydın et al., 2023; Isimekhai et al., 2017). Pakistan enjoys four seasons 400 i.e., dry autumn between September and November, dry and cold winter during December and 401 February, spring from March to May, warm and rainy summer which generally lasts from June to 402 403 August. Since the present study was carried out for one year covering all four seasons, seasonal comparisons of concentrations of studied HMs (Cr, Mn, Ni, Cu, Zn, Cd, Pb) in air and soil was 404 undertaken for each city. Seasonal mean deposition fluxes of HMs in air are presented in Table 405 406 S13 and their seasonal variations are illustrated in Figure 4a. On average, higher mean fluxes of all HMs in air were recorded either in Winter or Autumn whereas lowest fluxes were observed 407 during the summer except for Zn,. Mean deposition fluxes ($\mu g/m^2$.day) of Cu (122), Cd (2.4), and 408 Pb (159) were observed in Autumn whereas those for Cr (25.3), Mn (167) and Ni (67.8) were 409 found during winter. For Zn, higher mean deposition fluxes were inversely found during the spring 410 season. As presented in Table S13, this rise in fluxes was only contributed by elevated levels of Zn 411 in Karachi and Lahore. This might be due to the contribution by some additional sources at sites 412 and/or extraordinary dismantling / recycling. 413

Average seasonal concentrations of selected HMs in soil at sampling sites are given in Table S14. Similarly, higher concentrations of HMs in soil were associated with the dry season *i.e.*, autumn followed by winter, spring, and summer. Mean concentrations (in μ g/g dw) of Cr, Mn, Ni, Cu, Cd, Pb were recorded as 59.6, 415.2, 47.8, 188.2, 0.8, 118.5 respectively except for Zn for which the mean elevated levels were observed in spring likewise levels in air. In contrast to variability pattern as observed for air levels, high seasonal variability was observed for most of the HMs (Figure 4b). The spatial spread of HMs could be attributed to magnitude of e-waste dismantling / recycling

activities in addition to variable rainfall patterns in different cities, surface runoffs, human 421 activities across the sites and soil characteristics (Isimekhai et al., 2017). The range of HMs 422 423 concentrations at e-waste processing sites depends on nature of activities. Isimekhai et al., 2017 had associated clustering of Cd, Cu, Cr, Pb and Zn with recycling activities, whereas the presence 424 of Ni and Mn indicates dismantling activities. Nevertheless, no significant difference (p > 0.05) in 425 426 HMs concentrations were found in different seasons despite different meteorological conditions which shows that sampling sites are hot spots of HMs throughout the year. While positive 427 correlation (p < 0.05) of most of the heavy metals in soil and in Particulate matter suggest common 428 source of contamination (Table S15). 429

430 3.4. *Geo-accumulation index* (I_{geo})

Geo-accumulation index (Igeo) was determined based on the comparison between concentration 431 level of heavy metals at e-waste recycling facilities with the background site (Islamabad). The 432 calculated Igeo values for given recycling sites at sampling cities have been presented in 433 434 supplementary information (Table S16) and assessed with the given criteria for determining the scale of contamination. The I_{geo} values for each heavy metal was interpreted as follows: ≤ 0 435 (uncontaminated); $0 \le 1$ (uncontaminated – moderately contaminated); $1 \le 2$ (moderately 436 contaminated); $2 - \le 3$ (moderately – heavily contaminated); $3 - \le 4$ (heavily contaminated); 4 -437 \leq 5 (heavily to extremely contaminated) and 5 < (extremely contaminated). 438

Among sampling cities, Lahore, Gujranwala, Karachi, and Faisalabad were the most contaminated ones based on soil residues of the heavy metals. More specifically, Lahore and Gujranwala were extremely contaminated by Cu, while heavily to extremely contaminated by Pb and Cd, respectively. Similarly, Karachi and Faisalabad were marked for moderately to extremely contaminated for all target heavy metals especially Cu, Pb, Ni and Zn. Rawalpindi and Multan

have high Cu contamination levels, whereas these sites were moderately to heavily contaminated 444 with Zn, Cd, and Pb. Interestingly, Peshawar and Hyderabad showed moderate contamination of 445 all metals except Cd in Hyderabad. However, the Igeo levels for studied sites are elevated while 446 comparing Igeo levels of some other regional e-waste recycling locations of the world, e.g., informal 447 e-waste recycling shops in Dhaka, Bangladesh (Mowla et al., 2021) and inside the dumping area 448 449 of e-waste recycling facility at Korle Lagoon, Ghana for Ni, Pb and Cu (Fosu-Mensah et al., 2017). The present study found moderate to extreme levels of contamination of soil around e-waste 450 recycling locations (primarily in Pakistan's megacities) by most e-waste oriented heavy metals, 451 452 raising concerns about their possible exposure to workers and the surrounding environment.

453 **3.5.** *Enrichment Factor (EF)*

Table S17 presents the contamination factor results of HMs in soil and particulate samples of 40 454 recycling facilities throughout the country. The 455 e-waste mean EF value of 12.7>8.27>7.83>7.16>1.29>1.28>1.00 Pb>Zn>Cd>Cu>Ni>Cr>Mn was in soil, while 456 Zn>Cd>Pb>Cu>Ni>Cr>Mn was 78.9>78.5>64.6>15.2>4.89>1.49>1.00 for particulate samples 457 respectively. Among cities, higher EF values were calculated for Gujranwala, Lahore, Karachi, 458 Peshawar, and Quetta, being the most populated and industrialized cities depicting higher 459 460 contamination levels. In both sampling matrices, EF values indicate elevated contamination between e-waste recycling facilities for most of the metals studied except for Cr whereas Ni shows 461 a considerable contamination level. Pb, Cd, Zn and Cu levels were found to be elevated in 462 463 comparison with levels reported from e-waste recycling sites in India (Pradhan & Kumar, 2014). Similarly, higher levels of Cu, Pb, Cd and Zn and low levels of Mn and Ni were calculated from 464 465 another study conducted in India (Arya et al., 2021). The contamination levels at e-waste recycling

sites ranged from substantial or moderate contamination to extremely high contamination inPakistan's megacities.

468 **3.6.** *Human health risk assessment*

In soil ADI_{soil-ing} was the main exposure pathway to the workers in proximity of e-waste recycling 469 sites in all sampling cities (Table S18). Higher ADIsoil-ing was calculated for Zn, Mn, Pb and Cu, 470 while ADI_{soil-inh} was the least exposure pathway in all sampling cities. Non-CRs exposure of HQ 471 through different exposure routes suggests that HQsoil-der was the major route followed by HQsoil-472 ing and HQ_{soil-inh} in all sampling cities (Table S20). All elements did not pose any threat with value 473 of (HI < 1) for workers residing near e-waste recycling sites. Our results trends were similar to the 474 previous studies on non-CRs health risk assessment (Dutta et al., 2022; Han et al., 2018; Singh et 475 al., 2018). HQ_{soil-der} was observed to be the main exposure route of heavy metals with high values 476 of HI were calculated for Gujranwala, Lahore, Karachi, and Faisalabad with a value of 3.27×10⁻¹, 477 2.63×10^{-1} , 2.43×10^{-1} and 2.22×10^{-1} , respectively. 478

Estimation model of daily intake of particulate matter through different route suggest that 479 inhalation exposure (ECPM-inh) was the major pathway for all studied metals in all sampling cities 480 followed by ingestion, while dermal exposure have least observed values (Table S19). High 481 inhalation exposure ($\mu g/m^3$) was observed for Zn, Pb and Cu for Karachi (4.21×10^{-1} , 1.22×10^{-1} , 482 5.34×10⁻²), Lahore (4.16×10⁻¹, 2.51×10⁻², 2.64×10⁻²) and Gujranwala (3.62×10⁻¹, 5.64×10⁻², 483 7.62×10^{-2}), respectively. Non-CRs model provide evidence that the value of HQ_{PM-inh} was >1 for 484 Ni at Gujranwala (3.35) and Hyderabad (1.20), Mn at Karachi (1.38) and Faisalabad (~1) (Table 485 S21) indicate the potential for adverse health effects (USEPA, 2001). While HI values of HQ_{PM-inh} 486 was >1 for Gujranwala (4.63), Karachi (3.37), Hyderabad (2.21), Faisalabad (1.76), Lahore (1.54) 487 Multan (1.25) and Quetta (1.17) suggest the chance of occurrence of non-CRs effects to the 488

workers and public living near e-waste recycling facilities in these cities. Previously, Aziz et al. 489 (2022) also reported high level heavy metal (particulate) exposure via inhalation route among 490 ingestion and dermal at Makkah city in Saudia Arabia. To summarize, non-CRs assessments 491 indicate that dermal exposure is the major route of exposure to the contaminated soil, while 492 inhalation for particulate matter. Karachi, Gujranwala, Lahore, Faisalabad were the most 493 494 contaminated cities while Pb, Zn, Cu, Cd are key elements (exposure and contamination) in soil and particulate matter to the workers and general population residing near e-waste recycling sites 495 in Pakistan. 496

497 **3.7.** *Lifetime Cancer Risk*

Among all investigated heavy metals, IARC-2024 (The International Agency for Research on 498 Cancer) has categorized Ni Cr, and Cd as group-1 carcinogens while Pb lies in group-2A 499 carcinogens. Cancer risk over lifetime (ILCR) of Cr, Ni, Cd and Pb via ingestion, inhalation and 500 dermal contact were determined and shown in Table 2. Moderate to very low ILCR was observed 501 for metals through different exposure routes in all cities (ILCR_{Σ inh}: 9.76×10⁻⁸ at Peshawar to 502 ILCR_{Σ der}:1.02×10⁻⁴ at Lahore for Pb). ILCR_{Σ inh} and ILCR_{Σ Der} were most common exposure 503 pathways for Cr being the major contributor of Σ ILCR in all e-waste recycling sites. The 504 accumulative ILCR ranged: high from Multan (1.08×10^{-4}) and low at Peshawar (9.53×10^{-5}) for 505 inhalation pathways. Overall, cumulative ILCR model suggest that inhalation and dermal contact 506 507 are main exposure route depicting moderate to low CRs for workers at e-waste recycling facilities 508 in Pakistan. The children living near and adults working in the proximity of these e-waste recycling sites may encounter acute as well as chronic health effects due to continuous exposures to HMs 509 510 (Wu et al., 2019). Moreover, local population living nearby these uncontrolled informal e-waste 511 recycling practices (through secondary exposure) in the studied cities especially Karachi, Lahore

and Faisalabad could potentially suffer health related problems by e.g., liver, and vascular system
disorders, chronic kidney damage, irritation of upper respiratory tract due to chronic HMs exposure
(Grant et al., 2013). In conclusion, soil and particulate matter contamination from e-waste
recycling operations poses a potentially alarming risk of cancer and other health issues in Pakistan.

516 **4.** *Conclusions*

In the last decade, there has been a rise in e-waste recycling activities in Pakistan to recover 517 valuable metals by open burning, dismantling without safety measures, treatment with acid bath, 518 and refurbishment without safety measures. However, very few studies have been carried out 519 where hazardous emissions in environmental media are conducted to assess their magnitude and 520 health effects. This is the first comprehensive study, where passive air samplers were deployed at 521 522 40 e-waste recycling facilities across multiple cities (n=9) for a year-long continuous sampling of 523 heavy metals (Cr, Ni, Pb, Cu, Mn, Zn and Cd). Simultaneously, the soil samples were also taken 524 from the same sites, once in each season. Comparatively, the major urban centers i.e., Karachi, 525 Lahore, Gujranwala, and Faisalabad showed the highest levels of HMs in air and soil. 526 Concentrations of HM in air and soil were generally higher during dry and cold seasons as 527 compared to wet and hot seasons. E-waste facilities were shown to be hotspots of HMs leading to 528 concern for workers at these facilities as well as nearby populations. The results of Igeo index depict that Gujranwala, Lahore, Karachi, and Faisalabad are in range of moderately to extremely 529 contaminated for most the HMs studied especially Cu, Ni, Pb and Cd. The present informal e-530 531 waste dismantling and recycling practices in Pakistan highlights the need for stricter regulatory frameworks around e-waste and improvements to recycling practices and technologies. 532

Given the large-scale informal e-waste recycling operations in the country, further researchinvolving more detailed sampling, especially reference sites in every studied city would help in

forming a clearer spatial trend. Transect studies for air and soil would also help to delineate the 535 extent to which these hot spots of HMs exert effects on local populations and the environment. In 536 addition, exploring bioavailability of HMs in air (particulate matter) and soil may also be an 537 additional source to determine the potential harmful risks caused by synergistic effects by presence 538 of HMs in several environmental matrices. For instance, oxidative potential of airborne particulate 539 540 matter, has been shown to be driven by high levels of certain HMs such as Fe and Cu, among others. HMs associated with particulate matter, when inhaled, lead to the formation of reactive 541 oxygen species, which negatively impact health through damage of cardiovascular and respiratory 542 tissues (Cohen et al., 2015; Shahpoury et al., 2021). 543

Supplementary Information. Supplementary information (SI) contains two figure & twenty one
(21) tables.

546 Acknowledgements. Authors are thankful to Mr. Awais-ur-Rehman (CUI), Dr. Noor Muhammad

547 (CCRI, Multan), Dr. Naz Yasin Naz (University of Agriculture Faisalabad), Dr. Zoobia Khattak

548 (Benazir Women University, Peshawar), Dr. Naeem Akhtar Abbasi (University of Punjab, Lahore),

549 Mr. Taimoor Durrani (BUITEM) and Dr. Aman Ullah Mehr (University of Sindh) for providing

support in installation of passive samplers and collection of soil samples.

551 Funding.

- 552 This work has been supported by Pakistan Science Foundation (PSF) # Env/C-COMSATS Isb
- 553 (14) and The Scientific and Technological Research Institution of Türkiye (Grant # 118Y216).
- **Competing interests.** The authors declare no competing financial interest.

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Figure 1: Geographical locations of cities in focus and sampling network (refer to Table S1 for
which sampling site code (J1, J2 etc.) belongs to which sampling site)



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820 Figure 2: An intra-city comparison of heavy metals (flux (μg/m².day) in air



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822 Figure 3: An intra-city comparison of targeted metals in soils (μg/g dw)



Figure 4: Seasonal variations of studied heavy metals in (a). air & (b). soil

Table 1: Descriptive statistics of heavy metals at e-waste recycling facilities (all studied sites) and background site (Islamabad) in Pakistan

	Concentration level in air (Flux, µg/m².day)				Concentration level in soils (µg/g dw)			
UМс	E-waste Recycling sites		Background site		E-waste Recycling sites		Background site	
TIVIS	Mean+S.D	Range	Mean+S.D	Range	Mean+S.D	Range	Mean+S.D	Range
Cr	18.9±13.8	2.3-131	5.34±2.58	2.23-7.66	49.4±35.5	3.6-716	19.5±3.10	16.5-23.8
Mn	116±95.8	10.5-1520	33.47±13.3	15.8-45.8	372±152	25.1-1599	210±59.9	161-296
Ni	42.2±61.8	1.1-1057	10.0±7.33	3.08-20.2	37.3±26.1	2.9-542	12.2±4.60	8.3-18.8
Cu	97±52.4	7.5-1000	24.7±17.0	5.67-40.8	155±85.3	5.4-2784	11.9±2.06	8.9-13.63
Zn	733±273	23.6-8105	169±110	67.1-323	220±96.9	15.8-2301	45.5±3.81	41.6-49.3
Cd	1.8±0.3	0.1-23.4	0.18±0.09	0.03-0.24	0.6±0.3	0.03-6.3	0.11±0.02	0.10-0.14
Pb	112±74.5	5.9-2992	49.9±86.1	2.35-179	63.1±46.3	2.2-2786	7.97±2.07	4.87-9.16
Avg.	161±111	0.1-8105	41.8 + 33.8	0.03-179	128±63.2	0.03-2786	43.8±10.8	0.10-296

Sampling cities	Exposure Risk	Cr	Ni	Cd	Pb	ΣILCR
	$ILCR_{\Sigma ing}$	8.55×10 ⁻⁶	1.05×10 ⁻⁵	1.51×10 ⁻⁶	7.18×10 ⁻⁸	2.06×10 ⁻⁵
Peshawer	ILCR _{Σinh}	9.47×10 ⁻⁵	3.91×10 ⁻⁷	1.75×10 ⁻⁷	9.76×10 ⁻⁸	9.53×10 ⁻⁵
	$ILCR_{\Sigma Der}$	1.70×10 ⁻⁴	2.55×10 ⁻⁶	1.51×10 ⁻⁷	1.81×10 ⁻⁵	1.91×10 ⁻⁴
	ILCR _{Σing}	9.32×10 ⁻⁶	1.38×10 ⁻⁵	6.66×10 ⁻⁶	3.06×10 ⁻⁷	3.01×10 ⁻⁵
Faislabad	$ILCR_{\Sigma inh}$	1.91×10 ⁻⁴	7.51×10 ⁻⁷	9.49×10 ⁻⁸	5.33×10 ⁻⁸	1.92×10 ⁻⁴
	ILCR _{2Der}	1.86×10 ⁻⁴	3.35×10 ⁻⁶	6.79×10 ⁻⁷	7.48×10 ⁻⁵	2.64×10 ⁻⁴
	$ILCR_{\Sigma ing}$	7.47×10 ⁻⁶	8.41×10 ⁻⁶	2.05×10 ⁻⁶	1.06×10 ⁻⁷	1.80×10 ⁻⁵
Rawalpindi	$ILCR_{\Sigma inh}$	6.68×10 ⁻⁵	4.31×10 ⁻⁷	7.27×10 ⁻⁸	2.13×10 ⁻⁸	6.73×10 ⁻⁵
	$ILCR_{\Sigma Der}$	1.49×10 ⁻⁴	2.04×10 ⁻⁶	2.08×10 ⁻⁷	2.58×10-5	1.77×10 ⁻⁴
	$ILCR_{\Sigma ing}$	1.22×10 ⁻⁵	1.52×10 ⁻⁵	4.52×10 ⁻⁶	4.18×10 ⁻⁷	3.23×10 ⁻⁵
Lahore	$ILCR_{\Sigma inh}$	1.38×10 ⁻⁴	6.16×10 ⁻⁷	5.96×10 ⁻⁷	1.14×10 ⁻⁷	1.39×10 ⁻⁴
	$ILCR_{\Sigma Der}$	2.43×10 ⁻⁴	3.69×10 ⁻⁶	4.52×10 ⁻⁷	1.02×10 ⁻⁴	3.49×10 ⁻⁴
	$ILCR_{\Sigma ing}$	8.00×10 ⁻⁶	1.27×10 ⁻⁵	5.35×10 ⁻⁶	5.97×10 ⁻⁷	2.66×10 ⁻⁵
Karachi	$ILCR_{\Sigma inh}$	3.15×10 ⁻⁴	1.08×10 ⁻⁶	6.80×10 ⁻⁷	5.19×10 ⁻⁷	3.17×10 ⁻⁴
	$ILCR_{\Sigma Der}$	1.59×10 ⁻⁴	3.07×10 ⁻⁶	5.35×10 ⁻⁷	1.48×10 ⁻⁴	3.11×10 ⁻⁴
	$ILCR_{\Sigma ing}$	2.25×10-5	2.71×10 ⁻⁵	8.13×10 ⁻⁶	2.98×10-7	5.80×10 ⁻⁵
Gujranwala	$ILCR_{\Sigma inh}$	2.01×10 ⁻⁴	3.86×10 ⁻⁶	9.61×10 ⁻⁷	2.40×10 ⁻⁷	2.06×10 ⁻⁴
	$ILCR_{\Sigma Der}$	4.48×10-4	6.54×10 ⁻⁶	8.15×10 ⁻⁷	7.40×10 ⁻⁵	5.30×10 ⁻⁴
	$ILCR_{\Sigma ing}$	4.97×10 ⁻⁶	6.39×10 ⁻⁶	1.98×10 ⁻⁶	6.06×10 ⁻⁸	1.34×10 ⁻⁵
Multan	$ILCR_{\Sigma inh}$	1.07×10 ⁻⁴	4.64×10 ⁻⁷	1.09×10 ⁻⁷	8.15×10 ⁻⁸	1.08×10 ⁻⁴
	$ILCR_{\Sigma Der}$	9.89×10 ⁻⁵	1.55×10 ⁻⁶	2.01×10 ⁻⁷	1.52×10 ⁻⁵	1.16×10 ⁻⁴
	$ILCR_{\Sigma ing}$	8.94×10 ⁻⁶	1.26×10 ⁻⁵	9.13×10 ⁻⁷	4.14×10 ⁻⁸	2.25×10 ⁻⁵
Quetta	$ILCR_{\Sigma inh}$	1.16×10 ⁻⁴	3.75×10-7	1.61×10 ⁻⁷	1.05×10 ⁻⁷	1.16×10 ⁻⁴
	$ILCR_{\Sigma Der}$	1.78×10 ⁻⁴	3.06×10 ⁻⁶	9.05×10 ⁻⁸	1.07×10 ⁻⁵	1.92×10 ⁻⁴
	$ILCR_{\Sigma ing}$	7.76×10 ⁻⁶	8.14×10 ⁻⁶	4.25×10 ⁻⁶	6.25×10 ⁻⁸	2.02×10 ⁻⁵
Hyderabad	$ILCR_{\Sigma inh}$	1.62×10 ⁻⁴	1.39×10 ⁻⁶	1.21×10 ⁻⁷	5.15×10 ⁻⁸	1.63×10 ⁻⁴
	$ILCR_{\Sigma Der}$	1.55×10 ⁻⁴	1.96×10 ⁻⁶	4.33×10 ⁻⁷	1.55×10 ⁻⁵	1.72×10 ⁻⁴

Table 2: ILCRing, ILCRinh, ILCRdermal and Cumulative ILCR for Cd, Cr, Ni and Pb (carcinogenic heavy metals)